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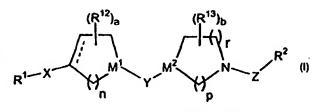
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#### (54) Title: (1-4-PIPERIDINYL) BENZIMIDAZOLE DERIVATIVES USEFUL AS HISTAMINE H3 ANTAGONISTS



(57) Abstract: Disclosed are histamine  $H_3$  antagonists of the formula (I) wherein  $R^1$  is optionally substituted benzimidazolyl or a derivative thereof;  $R^2$  is optionally substituted aryl or heteroaryl;  $M_1$  and  $M^2$  are  $C(R^3)$  or N; Q is  $-N(R^8)$ -, -S- or -O-; and the remaining variables are as defined in the specification. Also disclosed are pharmaceutical compositions comprising the compounds of formula I and methods of treating

various diseases or conditions, such as allergy, allergy-induced airway responses, and congestion (e.g., nasal congestion) using the compounds of formula I. Also disclosed are methods of treating said diseases or conditions using the compounds of formula I in combination with an  $H_1$  receptor antagonist.



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(1-4-PIPERIDINYL) BENZIMIDAZOLE DERIVATIVES USEFUL AS HISTAMINE H3 ANTAGONISTS

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#### FIELD OF THE INVENTION

The present invention relates to novel substituted benzimidazoles and aza- and diaza-derivatives thereof useful as histamine H<sub>3</sub> antagonists. The invention also relates to pharmaceutical compositions comprising said compounds and their use in treating inflammatory diseases, allergic conditions and central nervous system disorders. The invention also relates to the use of a combination of novel histamine H<sub>3</sub> antagonists of this invention with histamine H<sub>1</sub> compounds for the treatment of inflammatory diseases and allergic conditions, as well as pharmaceutical compositions comprising a combination of one or more novel histamine H<sub>3</sub> antagonist compounds of the invention with one or more histamine H<sub>1</sub> compounds.

#### BACKGROUND OF THE INVENTION

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The histamine receptors, H<sub>1</sub>, H<sub>2</sub> and H<sub>3</sub> are well-identified forms. The H<sub>1</sub> receptors are those that mediate the response antagonized by conventional antihistamines. H<sub>1</sub> receptors are present, for example, in the ileum, the skin, and the bronchial smooth muscle of humans and other mammals. Through H<sub>2</sub> receptor-mediated responses, histamine stimulates gastric acid secretion in mammals and the chronotropic effect in isolated mammalian atria.

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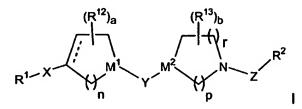
H<sub>3</sub> receptor sites are found on sympathetic nerves, where they modulate sympathetic neurotransmission and attenuate a variety of end organ responses under control of the sympathetic nervous system. Specifically, H<sub>3</sub> receptor activation by histamine attenuates norepinephrine outflow to resistance and capacitance vessels, causing vasodilation.

Imidazole H<sub>3</sub> receptor antagonists are well known in the art. More recently, non-imidazole H<sub>3</sub> receptor antagonists have been disclosed in PCT US01/32151, filed October 15, 2001, and US Provisional Application 60/275,417, filed March 13, 2001.

US 5,869,479 discloses compositions for the treatment of the symptoms of allergic rhinitis using a combination of at least one histamine H<sub>1</sub> receptor antagonist and at least one histamine H<sub>3</sub> receptor antagonist.

#### **SUMMARY OF THE INVENTION**

The present invention provides novel compounds of formula I:



or a pharmaceutically acceptable salt or solvate thereof, wherein:

the dotted line represents an optional double bond;

a is 0 to 2;

b is 0 to 2;

n is 1, 2 or 3;

p is 1, 2 or 3;

r is 0, 1, 2, or 3;

with the provisos that when  $M^2$  is N, p is not 1; and that when r is 0,  $M^2$  is  $C(R^3)$ ; and that the sum of p and r is 1 to 4;

 $M^1$  is  $C(R^3)$  or N;

20  $M^2$  is  $C(R^3)$  or N;

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X is a bond or C<sub>1</sub>-C<sub>6</sub> alkylene;

Y is -C(O)-, -C(S)-,  $-(CH_2)_q$ -,  $-NR^4C(O)$ -,  $-C(O)NR^4$ -,  $-C(O)CH_2$ -,  $-SO_2$ -,  $-N(R^4)$ -, -NH--C(=N-CN)- or -C(=N-CN)-NH-; with the provisos that when  $M^1$  is N, Y is not  $-NR^4C(O)$ - or -NH--C(=N-CN)-; when  $M^2$  is N, Y is not  $-C(O)NR^4$ - or -C(=N-CN)-NH-; and when Y is  $-N(R^4)$ -,  $M^1$  is CH and  $M^2$  is  $C(R^3)$ ;

q is 1 to 5, provided that when both M<sup>1</sup> and M<sup>2</sup> are N, q is 2 to 5;

Z is a bond,  $C_1$ - $C_6$  alkylene,  $C_1$ - $C_6$  alkenylene, -C(O)-, -CH(CN)-, - $SO_2$ - or - $CH_2C(O)NR^4$ -;

Q is -N(R<sup>8</sup>)-, -S- or -O-;
k is 0, 1, 2, 3 or 4;
k1 is 0, 1, 2 or 3;
k2 is 0, 1 or 2;

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R is H,  $C_1$ - $C_6$  alkyl, halo( $C_1$ - $C_6$ )alkyl-,  $C_1$ - $C_6$  alkoxy, ( $C_1$ - $C_6$ )alkoxy-

10 (C<sub>1</sub>-C<sub>6</sub>)alkyl-, (C<sub>1</sub>-C<sub>6</sub>)-alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkyl-SO<sub>0-2</sub>,  $R^{32}\text{-aryl}(C_1\text{-}C_6)\text{alkoxy-}, R^{32}\text{-aryl}(C_1\text{-}C_6)\text{alkyl-}, R^{32}\text{-aryl}, R^{32}\text{-aryloxy}, R^{32}\text{-heteroaryl}, \\ (C_3\text{-}C_6)\text{cycloalkyl}, (C_3\text{-}C_6)\text{cycloalkyl-}(C_1\text{-}C_6)\text{alkyl}, (C_3\text{-}C_6)\text{cycloalkyl-}(C_1\text{-}C_6)\text{alkoxy}, \\ (C_3\text{-}C_6)\text{cycloalkyl-oxy-}, R^{37}\text{-heterocycloalkyl}, R^{37}\text{-heterocycloalkyl-oxy-}, \\ R^{37}\text{-heterocycloalkyl-}(C_1\text{-}C_6)\text{alkoxy}, N(R^{30})(R^{31})\text{-}(C_1\text{-}C_6)\text{alkyl-}, -N(R^{30})(R^{31}), \\ R^{39}\text{-}(C_1\text{-}C_6)\text{-}(C_1\text{-}C_6$ 

15 -NH-(C<sub>1</sub>-C<sub>6</sub>)alkyl-O-(C<sub>1</sub>-C<sub>6</sub>)alkyl, -NHC(O)NH(R<sup>29</sup>);  $R^{29}$ -S(O)<sub>0-2</sub>-, halo(C<sub>1</sub>-C<sub>6</sub>)alkyl-S(O)<sub>0-2</sub>-, N(R<sup>30</sup>)(R<sup>31</sup>)-(C<sub>1</sub>-C<sub>6</sub>)alkyl-S(O)<sub>0-2</sub>- or benzoyl;

 $R^{8} \text{ is H, C}_{1}\text{-}C_{6} \text{ alkyl, halo}(C_{1}\text{-}C_{6}) \text{alkyl-, } (C_{1}\text{-}C_{6}) \text{alkoxy-}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{32}\text{-aryl}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{32}\text{-aryl, } R^{32}\text{-heteroaryl, } (C_{3}\text{-}C_{6}) \text{cycloalkyl, } (C_{3}\text{-}C_{6}) \text{cycloalkyl-}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{37}\text{-heterocycloalkyl, } N(R^{30})(R^{31})\text{-}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{29}\text{-}S(O)_{2}\text{-, halo}(C_{1}\text{-}C_{6}) \text{alkyl-}S(O)_{2}\text{-, } R^{29}\text{-}S(O)_{0}\text{-}_{1}\text{-}(C_{2}\text{-}C_{6}) \text{alkyl-, } halo(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{29}\text{-}S(O)_{0}\text{-}_{1}\text{-}(C_{2}\text{-}C_{6}) \text{alkyl-, } R^{29}\text{-}_{2}\text{-}(C_{2}\text{-}C_{6}) \text{alkyl-, } R$ 

R<sup>2</sup> is a six-membered heteroaryl ring having 1 or 2 heteroatoms independently selected from N or N-O, with the remaining ring atoms being carbon; a five-membered

heteroaryl ring having 1, 2, 3 or 4 heteroatoms independently selected from N, O or S, with the remaining ring atoms being carbon;  $R^{32}$ -quinolyl;  $R^{32}$ -aryl; heterocycloalkyl;  $(C_3-C_6)$ cycloalkyl;  $C_1-C_6$  alkyl; hydrogen; thianaphthenyl;

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—N  $\xi$ —N

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wherein said six-membered heteroaryl ring or said five-membered heteroaryl ring is optionally substituted by R<sup>6</sup>;

R<sup>3</sup> is H, halogen, C<sub>1</sub>-C<sub>6</sub> alkyl, -OH, (C<sub>1</sub>-C<sub>6</sub>)alkoxy or -NHSO<sub>2</sub>-(C<sub>1</sub>-C<sub>6</sub>)alkyl;

 $R^4$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_6$  alkyl,  $C_3$ - $C_6$  cycloalkyl,  $(C_3$ - $C_6)$ cycloalkyl,  $(C_1$ - $C_6)$ alkyl,  $R^{33}$ -aryl,  $R^{33}$ -ar

 $R^5$  is hydrogen,  $C_1$ - $C_6$  alkyl,  $-C(O)R^{20}$ ,  $-C(O)_2R^{20}$ ,  $-C(O)N(R^{20})_2$ ,  $(C_1$ - $C_6)$ alkyl- $SO_2$ -, or  $(C_1$ - $C_6)$ alkyl- $SO_2$ -NH-;

or R<sup>4</sup> and R<sup>5</sup>, together with the nitrogen to which they are attached, form an azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl or morpholinyl ring;

R<sup>6</sup> is 1 to 3 substituents independently selected from the group consisting of -OH, halogen, C<sub>1</sub>-C<sub>6</sub> alkyl-, C<sub>1</sub>-C<sub>6</sub> alkoxy, C<sub>1</sub>-C<sub>6</sub> alkylthio, -CF<sub>3</sub>, -NR<sup>4</sup>R<sup>5</sup>, -CH<sub>2</sub>-NR<sup>4</sup>R<sup>5</sup>, -NHSO<sub>2</sub>R<sup>22</sup>, -N(SO<sub>2</sub>R<sup>22</sup>)<sub>2</sub>, phenyl, R<sup>33</sup>-phenyl, NO<sub>2</sub>, -CO<sub>2</sub>R<sup>4</sup>, -CON(R<sup>4</sup>)<sub>2</sub>,

 $R^7$  is -N( $R^{29}$ )-, -O- or -S(O)<sub>0-2</sub>-;

 $R^{12}$  is independently selected from the group consisting of  $C_1$ - $C_6$  alkyl, hydroxyl,  $C_1$ - $C_6$  alkoxy, or fluoro, provided that when  $R^{12}$  is hydroxy or fluoro, then  $R^{12}$  is not bound to a carbon adjacent to a nitrogen; or two  $R^{12}$  substituents form a  $C_1$  to  $C_2$  alkyl bridge from one ring carbon to another non-adjacent ring carbon; or  $R^{12}$  is =0;

 $R^{13}$  is independently selected from the group consisting of  $C_1$ - $C_6$  alkyl, hydroxyl,  $C_1$ - $C_6$  alkoxy, or fluoro, provided that when  $R^{13}$  is hydroxy or fluoro then  $R^{13}$  is not bound to a carbon adjacent to a nitrogen; or two  $R^{13}$  substituents form a  $C_1$  to  $C_2$  alkyl bridge from one ring carbon to another non-adjacent ring carbon; or  $R^{13}$  is =O;

R<sup>20</sup> is independently selected from the group consisting of hydrogen, C<sub>1</sub>-C<sub>6</sub> alkyl, or aryl, wherein said aryl group is optionally substituted with from 1 to 3 groups independently selected from halogen, -CF<sub>3</sub>, -OCF<sub>3</sub>, hydroxyl, or methoxy; or when two

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R<sup>20</sup> groups are present, said two R<sup>20</sup> groups taken together with the nitrogen to which they are bound can form a five or six membered heterocyclic ring;

R<sup>22</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl, R<sup>34</sup>-aryl or heterocycloalkyl;

 $R^{24}$  is H, C<sub>1</sub>-C<sub>6</sub> alkyl, -SO<sub>2</sub>R<sup>22</sup> or R<sup>34</sup>-aryl;

 $R^{25}$  is independently selected from the group consisting of  $C_1$ - $C_6$  alkyl, halogen, -CN, -NO<sub>2</sub>, -CF<sub>3</sub>, -OH, C<sub>1</sub>-C<sub>6</sub> alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkyl-C(O)-, aryl-C(O)-, -C(O)OR<sup>29</sup>, -N(R<sup>4</sup>)(R<sup>5</sup>), N(R<sup>4</sup>)(R<sup>5</sup>)-C(O)-, N(R<sup>4</sup>)(R<sup>5</sup>)-S(O)<sub>1-2</sub>-, R<sup>22</sup>-S(O)<sub>0-2</sub>-, halo-(C<sub>1</sub>-C<sub>6</sub>)alkyl- or halo-(C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkyl-;

 $R^{29}$  is H,  $C_1$ - $C_6$  alkyl,  $C_3$ - $C_6$  cycloalkyl,  $R^{35}$ -aryl or  $R^{35}$ -aryl( $C_1$ - $C_6$ )alkyl-;

 $R^{30}$  is H,  $C_1$ - $C_6$  alkyl-,  $R^{35}$ -aryl or  $R^{35}$ -aryl( $C_1$ - $C_6$ )alkyl-;

 $R^{31}$  is H,  $C_1$ - $C_6$  alkyl-,  $R^{35}$ -aryl,  $R^{35}$ -aryl( $C_1$ - $C_6$ )alkyl-,  $R^{35}$ -heteroaryl, ( $C_1$ - $C_6$ )alkyl-C(O)-,  $R^{35}$ -aryl-C(O)-,  $N(R^4)(R^5)$ -C(O)-, ( $C_1$ - $C_6$ )alkyl-S(O)<sub>2</sub>- or  $R^{35}$ -aryl-S(O)<sub>2</sub>-;

or  $R^{30}$  and  $R^{31}$  together are -(CH<sub>2</sub>)<sub>4-5</sub>-, -(CH<sub>2</sub>)<sub>2</sub>-O-(CH<sub>2</sub>)<sub>2</sub>- or

 $-(CH_2)_2-N(R^{38})-(CH_2)_2-$  and form a ring with the nitrogen to which they are attached;

 $R^{32}$  is 1 to 3 substituents independently selected from the group consisting of H, -OH, halogen,  $C_1$ - $C_6$  alkyl,  $C_1$ - $C_6$  alkoxy,  $R^{35}$ -aryl-O-, -S $R^{22}$ , -CF<sub>3</sub>, -OCF<sub>3</sub>, -OCHF<sub>2</sub>, -NR<sup>39</sup>R<sup>40</sup>, phenyl,  $R^{33}$ -phenyl,  $NO_2$ , -CO<sub>2</sub>R<sup>39</sup>, -CON( $R^{39}$ )<sub>2</sub>, -S(O)<sub>2</sub>R<sup>22</sup>, -S(O)<sub>2</sub>N( $R^{20}$ )<sub>2</sub>, -N( $R^{24}$ )S(O)<sub>2</sub>R<sup>22</sup>, -CN, hydroxy-( $C_1$ - $C_6$ )alkyl-, -OCH<sub>2</sub>CH<sub>2</sub>OR<sup>22</sup>, and

R<sup>35</sup>-aryl(C<sub>1</sub>-C<sub>6</sub>)alkyl-O-, or two R<sup>32</sup> groups on adjacent carbon atoms together form a –OCH<sub>2</sub>O- or –O(CH<sub>2</sub>)<sub>2</sub>O- group;

R<sup>33</sup> is 1 to 3 substituents independently selected from the group consisting of C<sub>1</sub>-C<sub>6</sub> alkyl, halogen, -CN, -NO<sub>2</sub>, -CF<sub>3</sub>, -OCF<sub>3</sub>, -OCHF<sub>2</sub> and -O-(C<sub>1</sub>-C<sub>6</sub>)alkyl;

R<sup>34</sup> is 1 to 3 substituents independently selected from the group consisting of H, halogen, -CF<sub>3</sub>, -OCF<sub>3</sub>, -OH and -OCH<sub>3</sub>;

R<sup>35</sup> is 1 to 3 substituents independently selected from hydrogen, halo, C<sub>1</sub>-C<sub>6</sub> alkyl, hydroxy, C<sub>1</sub>-C<sub>6</sub> alkoxy, phenoxy, -CF<sub>3</sub>, -N(R<sup>36</sup>)<sub>2</sub>, -COOR<sup>20</sup> and -NO<sub>2</sub>;

R<sup>36</sup> is independently selected form the group consisting of H and C<sub>1</sub>-C<sub>6</sub> alkyl; R<sup>37</sup> is 1 to 3 substituents independently selected from hydrogen, halo, C<sub>1</sub>-C<sub>6</sub> alkyl, hydroxy, C<sub>1</sub>-C<sub>6</sub> alkoxy, phenoxy, -CF<sub>3</sub>, -N(R<sup>36</sup>)<sub>2</sub>, -COOR<sup>20</sup>, -C(O)N(R<sup>29</sup>)<sub>2</sub> and

30  $-NO_2$ , or  $R^{37}$  is one or two =O groups;

 $R^{36}$  is H,  $C_1$ - $C_6$  alkyl,  $R^{35}$ -aryl,  $R^{35}$ -aryl( $C_1$ - $C_6$ )alkyl-, ( $C_1$ - $C_6$ )alkyl-SO<sub>2</sub> or halo( $C_1$ - $C_6$ )alkyl-SO<sub>2</sub>-;

 $R^{39}$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_6$  alkyl,  $C_3$ - $C_6$  cycloalkyl,  $(C_3$ - $C_6$ )cycloalkyl( $C_1$ - $C_6$ )alkyl,  $R^{33}$ -aryl,  $R^{33}$ -aryl,  $R^{33}$ -aryl,  $R^{33}$ -aryl, and  $R^{32}$ -heteroaryl; and

 $R^{40}$  is hydrogen,  $C_1$ - $C_6$  alkyl,  $-C(O)R^{20}$ ,  $-C(O)_2R^{20}$ ,  $-C(O)N(R^{20})_2$ ,  $(C_1$ - $C_6)$ alkyl- $SO_2$ -, or  $(C_1$ - $C_6)$ alkyl- $SO_2$ -NH-;

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or R<sup>39</sup> and R<sup>40</sup>, together with the nitrogen to which they are attached, form an azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl or morpholinyl ring;

This invention also provides a pharmaceutical composition comprising an effective amount of compound of at least one compound of formula I and a pharmaceutically acceptable carrier.

This invention further provides a method of treating: allergy, allergy-induced airway (e.g., upper airway) responses, congestion (e.g., nasal congestion), hypotension, cardiovascular disease, diseases of the GI tract, hyper and hypo motility and acidic secretion of the gastro-intestinal tract, obesity, sleeping disorders (e.g., hypersomnia, somnolence, and narcolepsy), disturbances of the central nervous system, attention deficit hyperactivity disorder (ADHD), hypo and hyperactivity of the central nervous system (for example, agitation and depression), and/or other CNS disorders (such as Alzheimer's, schizophrenia, and migraine) comprising administering to a patient in need of such treatment (e.g., a mammal, such as a human being) an effective amount of at least one compound of formula I.

Compounds of this invention are particularly useful for treating allergy, allergy-induced airway responses and/or congestion.

This invention further provides a pharmaceutical composition comprising an effective amount of a combination of at least one compound of formula I and at least one H<sub>1</sub> receptor antagonist in combination with a pharmaceutically acceptable carrier.

This invention further provides a method of treating allergy, allergy-induced airway (e.g., upper airway) responses, and/or congestion (e.g., nasal congestion) comprising administering to a patient in need of such treatment (e.g., a mammal, such as a human being) an effective amount of a combination of at least one compound of formula I and at least one H<sub>1</sub> receptor antagonist.

Kits comprising a compound of formula I in a pharmaceutical composition, and a separate H<sub>1</sub> receptor antagonist in a pharmaceutical compositions in a single package are also contemplated.

#### **DETAILED DESCRIPTION OF THE INVENTION**

Preferred definitions of the variables in the structure of formula I are as follows:

R<sup>1</sup> is preferably optionally substituted benzimidazolyl or 7-azabenzimidazolyl, wherein R is preferably alkyl, alkoxy, alkoxyalkoxy, alkylthio, heteroaryl or R<sup>32</sup>-aryl. More preferably, R is –CH<sub>3</sub>, -CH<sub>2</sub>CH<sub>3</sub>, –OCH<sub>3</sub>, -OCH<sub>2</sub>CH<sub>3</sub>, -OCH<sub>2</sub>CH<sub>3</sub>, -OCH<sub>2</sub>CH<sub>3</sub>, -OCH<sub>4</sub>((CH<sub>3</sub>)<sub>2</sub>, -SCH<sub>3</sub>, -SCH<sub>2</sub>CH<sub>3</sub>, pyridyl (especially 2-pyridyl), pyrimidyl, pyrazinyl, furanyl, oxazolyl or R<sup>32</sup>-phenyl.

R<sup>25</sup> is preferably halogen or –CF<sub>3</sub> and k is 0 or 1.

 $R^2$  is preferably a six-membered heteroaryl ring, optionally substituted with one substituent. More preferably,  $R^2$  is pyrimidyl,  $R^6$ -pyrimidyl, pyridyl,  $R^6$ -pyridyl or pyridazinyl, wherein  $R^6$  is  $-NR^4R^5$ , wherein  $R^4$  and  $R^5$  are independently selected from the group consisting of H and  $(C_1-C_6)$ alkyl, or  $R^4$  and  $R^5$  together with the nitrogen to which they are attached form a pyrrolidinyl, piperidinyl or morpholinyl ring. More preferably,  $R^6$  is  $-NH_2$ .

X is preferably a bond.

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Y is preferably -C(O)-.

Z is preferably straight or branched C<sub>1</sub>-C<sub>3</sub> alkyl.

M<sup>1</sup> is preferably N; a is preferably 0; and n is preferably 2; the optional double bond is preferably not present (i.e., a single bond is present).

M<sup>2</sup> is preferably C(R<sup>3</sup>) wherein R<sup>3</sup> is hydrogen or fluorine; b is preferably 0; r is preferably 1; and p is preferably 2.

As used herein, the following terms have the following meanings, unless indicated otherwise:

alkyl (including, for example, the alkyl portions of arylalkyl and alkoxy) represents straight and branched carbon chains and contains from one to six carbon atoms;

alkylene represents a divalent straight or branched alkyl chain, e.g., ethylene (-CH<sub>2</sub>CH<sub>2</sub>-) or propylene (-CH<sub>2</sub>CH<sub>2</sub>-);

Haloalkyl and haloalkoxy represent alkyl or alkoxy chains wherein one or more hydrogen atoms are replaced by halogen atoms, e.g., -CF<sub>3</sub>, CF<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>-, CF<sub>3</sub>CF<sub>2</sub>- or CF<sub>3</sub>S;

aryl (including the aryl portion of arylalkyl) represents a carbocyclic group containing from 6 to 14 carbon atoms and having at least one aromatic ring (e.g., aryl

is a phenyl or naphthyl ring), with all available substitutable carbon atoms of the carbocyclic group being intended as possible points of attachment;

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arylalkyl represents an aryl group, as defined above, bound to an alkyl group, as defined above, wherein said alkyl group is bound to the compound;

cycloalkyl represents saturated carbocyclic rings of from 3 to 6 carbon atoms; halogen (halo) represents fluoro, chloro, bromo and iodo;

heteroaryl represents cyclic groups, having 1 to 4 heteroatoms selected from O, S or N, said heteroatom interrupting a carbocyclic ring structure and having a sufficient number of delocalized pi electrons to provide aromatic character, with the aromatic heterocyclic groups preferably containing from 2 to 14 carbon atoms; examples include but are not limited to isothiazolyl, isoxazolyl, oxazolyl, furazanyl, triazolyl, tetrazolyl, thiazolyl, thiadiazolyl, isothiadiazolyl, thienyl, furanyl (furyl), pyrrolyl, pyrazolyl, pyranyl, pyrimidinyl, pyrazinyl, pyridazinyl, pyridyl (e.g., 2-, 3-, or 4-pyridyl N-oxide), triazinyl, pteridinyl, indolyl (benzopyrrolyl), pyridopyrazinyl, isoqinolinyl, quinolinyl, naphthyridinyl; the 5- and 6-membered heteroaryl groups included in the definition of R<sup>2</sup> are exemplified by the heteroaryl groups listed above; all available substitutable carbon and nitrogen atoms can be substituted as defined;

heterocycloalkyl represents a saturated, carbocylic ring containing from 3 to 15 carbon atoms, preferably from 4 to 6 carbon atoms; examples include but are not limited to 2- or 3-tetrahydrofuranyl, 2- or 3- tetrahydrothienyl, 2-, 3- or 4-piperidinyl, 2- or 3-pyrrolidinyl, 2- or 3-piperazinyl, 2- or 4-dioxanyl, 1,3-dioxolanyl, 1,3,5-trithianyl, pentamethylene sulfide, perhydroisoquinolinyl, decahydroquinolinyl, trimethylene oxide, azetidinyl, 1-azacycloheptanyl, 1,3-dithianyl, 1,3,5-trioxanyl, morpholinyl, thiomorpholinyl, 1,4-thioxanyl, and 1,3,5-hexahydrotriazinyl, thiazolidinyl, tetrahydropyranyl.

In the definition of  $R^{32}$ , when two  $R^{32}$  groups on adjacent carbon atoms of an aryl or heteroaryl ring are said to be taken together form a  $-OCH_2O$ - or  $-O(CH_2)_2O$ -group, this means that the two  $R^{32}$  groups form a methylenedioxy or ethylenedioxy ring fused to the aryl or heteroaryl ring. When  $R^{12}$ ,  $R^{13}$  or  $R^{37}$  is said to be one or two =O groups, this means that two hydrogen atoms on the same carbon atom of the ring can be replaced by =O; two such groups can be present on a ring.

N , for example in the structure

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represents a nitrogen atom that is located at one of the 4 non-fused positions of the ring, i.e., positions 4, 5, 6 or 7 indicated below:

-9-

$$\begin{array}{c|c}
R & 1 \\
3 & N & 7 \\
4 & N & 7 \\
5 & 6
\end{array}$$

Similarly, means that two nitrogens are located at any two of the 4 non-fused positions of the ring, e.g., the 4 and 6 positions, the 4 and 7 positions, or the 5 and 6 positions.

Also, as used herein, "upper airway" usually means the upper respiratory system--i.e., the nose, throat, and associated structures.

Also, as used herein, "effective amount" generally means a therapeutically effective amount.

"Patient" means a mammal, typically a human, although veterinary use is also contemplated.

Lines drawn into the rings indicate that the indicated bond may be attached to any of the substitutable ring carbon atoms.

Certain compounds of the invention may exist in different isomeric (e.g., enantiomeric, diastereoisomeric and geometric) forms. The invention contemplates all such isomers both in pure form and in admixture, including racemic mixtures. Enol forms and tautomers are also included.

The compounds of this invention are ligands for the histamine  $H_3$  receptor. The compounds of this invention can also be described as antagonists of the  $H_3$  receptor, or as  $H_3$  antagonists.

The compounds of the invention are basic and form pharmaceutically acceptable salts with organic and inorganic acids. Examples of suitable acids for such salt formation are hydrochloric, sulfuric, phosphoric, acetic, citric, oxalic, malonic, salicylic, malic, fumaric, succinic, ascorbic, maleic, methanesulfonic and other mineral

and carboxylic acids well known to those skilled in the art. The salts are prepared by contacting the free base form with a sufficient amount of the desired acid to produce a salt in the conventional manner. The free base forms may be regenerated by treating the salt with a suitable dilute aqueous base solution such as dilute aqueous sodium hydroxide, potassium carbonate, ammonia and sodium bicarbonate. The free base forms differ from their corresponding salt forms somewhat in certain physical properties, such as solubility in polar solvents, but the salts are otherwise equivalent to their corresponding free base forms for purposes of this invention.

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Depending upon the substituents on the inventive compounds, one may be able to form salts with bases. Thus, for example, if there are carboxylic acid substituents in the molecule, salts may be formed with inorganic as well as organic bases such as, for example, NaOH, KOH, NH<sub>4</sub>OH, tetraalkylammonium hydroxide, and the like.

The compounds of formula I can exist in unsolvated as well as solvated forms, including hydrated forms, e.g., hemi-hydrate. In general, the solvated forms, with pharmaceutically acceptable solvents such as water, ethanol and the like are equivalent to the unsolvated forms for purposes of the invention.

The compounds of this invention can be combined with an H<sub>1</sub> receptor antagonist (i.e., the compounds of this invention can be combined with an H<sub>1</sub> receptor antagonist in a pharmaceutical composition, or the compounds of this invention can be administered with H<sub>1</sub> receptor antagonist).

Numerous chemical substances are known to have histamine H<sub>1</sub> receptor antagonist activity and can therefore be used in the methods of this invention. Many H<sub>1</sub> receptor antagonists useful in the methods of this invention can be classified as ethanolamines, ethylenediamines, alkylamines, phenothiazines or piperidines. Representative H<sub>1</sub> receptor antagonists include, without limitation: astemizole, azatadine, azelastine, acrivastine, brompheniramine, cetirizine, chlorpheniramine, clemastine, cyclizine, carebastine, cyproheptadine, carbinoxamine, descarboethoxyloratadine, diphenhydramine, doxylamine, dimethindene, ebastine, epinastine, efletirizine, fexofenadine, hydroxyzine, ketotifen, loratadine, levocabastine, meclizine, mizolastine, mequitazine, mianserin, noberastine, norastemizole, picumast, pyrilamine, promethazine, terfenadine, tripelennamine, temelastine, trimeprazine and triprolidine. Other compounds can readily be evaluated to determine activity at H<sub>1</sub> receptors by known methods, including specific blockade of the contractile response

to histamine of isolated guinea pig ileum. See for example, WO98/06394 published February 19, 1998.

Those skilled in the art will appreciate that the H<sub>1</sub> receptor antagonist is used at its known therapeutically effective dose, or the H<sub>1</sub> receptor antagonist is used at its normally prescribed dosage.

Preferably, said H<sub>1</sub> receptor antagonist is selected from: astemizole, azatadine, azelastine, acrivastine, brompheniramine, cetirizine, chlorpheniramine, clemastine, cyclizine, carebastine, cyproheptadine, carbinoxamine, descarboethoxyloratadine, diphenhydramine, doxylamine, dimethindene, ebastine, epinastine, efletirizine, fexofenadine, hydroxyzine, ketotifen, loratadine, levocabastine, meclizine, mizolastine, mequitazine, mianserin, noberastine, norastemizole, picumast, pyrilamine, promethazine, terfenadine, tripelennamine, temelastine, trimeprazine or triprolidine.

More preferably, said H<sub>1</sub> receptor antagonist is selected from: astemizole, azatadine, azelastine, brompheniramine, cetirizine, chlorpheniramine, clemastine, carebastine, descarboethoxyloratadine, diphenhydramine, doxylamine, ebastine, fexofenadine, loratadine, levocabastine, mizolastine, norastemizole, or terfenadine.

Most preferably, said H<sub>1</sub> receptor antagonist is selected from: azatadine, brompheniramine, cetirizine, chlorpheniramine, carebastine, descarboethoxy-loratadine, diphenhydramine, ebastine, fexofenadine, loratadine, or norastemizole.

Even more preferably, said H<sub>1</sub> antagonist is selected from loratadine, descarboethoxyloratadine, fexofenadine or cetirizine. Still even more preferably, said H<sub>1</sub> antagonist is loratadine or descarboethoxyloratadine.

In one preferred embodiment, said  $H_1$  receptor antagonist is loratedine. In another preferred embodiment, said  $H_1$  receptor antagonist is

25 descarboethoxyloratadine.

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In still another preferred embodiment, said H<sub>1</sub> receptor antagonist is fexofenadine.

In yet another preferred embodiment, said H<sub>1</sub> receptor antagonist is cetirizine. Preferably, in the above methods, allergy-induced airway responses are treated.

Also, preferably, in the above methods, allergy is treated.

Also, preferably, in the above methods, nasal congestion is treated.

In the methods of this invention wherein a combination of an  $H_3$  antagonist of this invention (compound of formula I) is administered with a  $H_1$  antagonist, the

antagonists can be administered simultaneously or sequentially (first one and then the other over a period of time). In general, when the antagonists are administered sequentially, the H<sub>3</sub> antagonist of this invention (compound of formula I) is administered first.

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Compounds of the present invention can be prepared by a number of ways evident to one skilled in the art. Preferred methods include, but are not limited to, the general synthetic procedures described herein. One skilled in the art will recognize that one route will be optimal depending on the choice of appendage substituents. Additionally, one skilled in the art will recognize that in some cases the order of steps has to be controlled to avoid functional group incompatibilities.

The starting material and reagents used in preparing compounds described are either available from commercial suppliers such as Aldrich Chemical Co. (Wisconsin, USA) and Acros Organics Co. (New Jersey, USA) or were prepared by literature methods known to those skilled in the art.

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One skilled in the art will recognize that the synthesis of compounds of formula I may require the construction of carbon-nitrogen bond. Methods include but are not limited to the use of a substituted aromatic compound or heteroaromatic compound and amine at 0 °C to 200 °C. The reaction may be carried out neat or in a solvent. Suitable solvents for the reaction are halogenated hydrocarbons, ethereal solvents, toluene, dimethylformamide and the like.

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One skilled in the art will recognize that the synthesis of compounds of formula I may require the construction of heterocycle. Methods include but are not limited to the use of a diamino compound and a carbonyl equivalent at 0 °C to 200 °C. The reaction may be carried out in acidic, basic or neutral conditions. Suitable solvents for the reaction are water, halogenated hydrocarbons, ethereal solvents, alcoholic solvents, toluene, ketones, dimethylformamide and the like.

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One skilled in the art will recognize that the synthesis of compounds of formula I may require the need for the protection of certain functional groups (i.e. derivatization for the purpose of chemical compatibility with a particular reaction condition). See, for example, Green et al, *Protective Groups in Organic Synthesis*. A suitable protecting group for an amine is methyl, benzyl, ethoxyethyl, t-butoxycarbonyl, phthaloyl and the like which can appended to and removed by literature methods known to those skilled in the art.

One skilled in the art will recognize that the synthesis of compounds of formula I may require the construction of an amide bond. Methods include but are not limited to the use of a reactive carboxy derivative (e.g. acid halide) or the use of an acid with a coupling reagent (e.g. EDCI, DCC, HATU) with an amine at 0 °C to 100 °C. Suitable solvents for the reaction are halogenated hydrocarbons, ethereal solvents, dimethylformamide and alike.

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One skilled in the art will recognize that the synthesis of compounds of formula I may require the reduction of a functional group. Suitable reducing reagents for the reaction include NaBH<sub>4</sub>, lithium aluminum hydride, diborane and the like at -20 °C to 100 °C. Suitable solvents for the reaction are halogenated hydrocarbons, ethereal solvents, and the like.

The starting materials and the intermediates of the reaction may be isolated and purified if desired using conventional techniques, including but not limited to filtration, distillation, crystallization, chromatography and alike. Such materials can be characterized using conventional means, including physical constants and spectral data.

One method shown in Scheme 1, below, is for the preparation of compounds of formula IA wherein R<sup>1</sup> is 1-benzimidazolyl or 2-benzamidazolyl and X is a bond or alkyl. Similar procedures can be used to prepare compounds wherein the benzene ring of the benzimidazolyl group is substituted, as well as the aza-benzimidazoles compounds (i.e., compounds wherein R<sup>1</sup> is other than benzimidazolyl as defined above) and the benzoxazolyl and benzothiazolyl derivatives.

SCHEME 1.

Step a: A suitably monoprotected diamine of formula X, wherein X is a bond or alkyl, Prot is a protecting group, and the remaining variables are as defined above is alkylated or arylated with a halide. The intermediate diamine is then cyclized with an appropriate carbonyl or formyl equivalent to form a compound of formula XI. Suitable

protecting groups are methyl, benzyl, butoxycarbonyl, or ethoxycarbonyl. A suitable halide for alkylation is a substituted aromatic compound or a substituted heteroaromatic compound as described by Henning et al, *J. Med. Chem. 30*, (1987), 814-819.

Step b: The protected amine of formula XI is deprotected using methods known to those skilled in the art. A suitable method for methyl deprotection is reaction with a haloformate or the like. A suitable method for benzyl deprotection is cleavage with hydrogen at or above atmospheric pressure and a catalyst such as palladium. Suitable methods for carbamate deprotection are treatment with an acid, base or trimethylsilyl iodide.

Step c: An amine of formula XII is reacted with an activated functional group Y of formula XIII to form the bond between the nitrogen and functional group Y in formula IA. When Y is a carbonyl group and M<sup>2</sup> is carbon, activation can be via a halide (i.e. acid chloride intermediate) or other coupling reagents (EDCI, DCC, HATU, or like). Suitable reaction conditions may require a base such as triethylamine or N,N-diisopropylethylamine.

Another method for the preparation of compounds of formula IA wherein R<sup>1</sup> is 1-benzimidazolyl or 2-benzimidazolyl and X is a bond or alkyl is shown in Scheme 2, below. Similar procedures can be used to prepare compounds wherein the benzene ring of the benzimidazolyl group is substituted, as well as the aza-benzimidazoles compounds (i.e., compounds wherein R<sup>1</sup> is other than benzimidazolyl as defined above).

#### Scheme 2.

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Step d: A suitably monoprotected diamine of formula X, wherein X is a bond or alkyl, Prot is a protecting group, and the remaining variables are as defined above, is alkylated or arylated with a halide to form a compound of formula XIV. Suitable protecting groups are methyl, benzyl, butoxycarbonyl, and ethoxycarbonyl. A suitable halide for alkylation is a substituted aromatic compound or a substituted heteroaromatic compound as described by Henning et al.

#### Step e:

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- (1) The protected amine of formula XIV is deprotected using methods known to those skilled in the art. A suitable method for methyl deprotection is reaction with a haloformate or the like. A suitable method for benzyl deprotection is cleavage with hydrogen at or above atmospheric pressure and a catalyst such as palladium. Suitable methods for carbamate deprotection are treatment with an acid, base or trimethylsilyl iodide.
- (2) The resulting amine from Step e(1) is reacted with an activated functional group Y of formula XIII to form the bond between the nitrogen and functional group Y to obtain the compound of formula XV. When Y is a carbonyl group and M² is carbon, activation can be via a halide (i.e. acid chloride intermediate) or other coupling reagents (EDCI, DCC, HATU, or the like). Suitable reaction conditions may require a base such as triethylamine, N,N-diisopropylethylamine, pyridine, or the like. <a href="Step f">Step f</a>: After reduction of formula XV, the resulting compound is reacted with a carbonyl equivalent to give the cyclized compound of formula IA. The reduction conditions can be hydrogen in the presence of catalyst, metal in the presence of an acid or a base, or other reduction reagent. The cyclization can be performed in acidic or basic conditions.

More detailed methods for synthesis of compounds are shown in **Scheme 3** below. The preparation of compounds of formula IB wherein R<sup>1</sup> is 1-benzimidazolyl (Methods A, B, C and F), Y is –C(O)- and R<sup>2</sup> is substituted pyridyl, and compounds of formulas IC and IC' wherein R<sup>1</sup> is 2-benzimidazolyl (Methods D and E), Y is –C(O)- and R<sup>2</sup> is substituted pyridyl are shown, but those skilled in the art will recognize that similar procedures can be used to prepare compounds wherein the benzene ring of the benzimidazolyl group is substituted, R<sup>2</sup> is other than pyridyl, and azabenzimidazoles compounds (i.e., compounds wherein R<sup>1</sup> is other than benzimidazolyl as defined above).

## Scheme 3.

Method A:

Method B:

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1 Step 1 NO2 NH 6 Step 2 O2N N 
$$M^2$$
 N-Z  $M$  R<sup>6</sup> (Prot)

9 Step 3 HN  $M^2$  N-Z  $M$  R<sup>6</sup> (Prot)

10 Step 4 HN  $M^2$  N-Z  $M$  R<sup>6</sup> (Prot)

10 Step 4 HN  $M^2$  N-Z  $M$  R<sup>6</sup> (Prot)

10 Step 4 HN  $M^2$  N-Z  $M$  R<sup>6</sup> (Prot)

Method C:

## Method E:

10 <u>Method F</u>:

Specifically exemplified compounds were prepared as described in the examples below, from starting materials known in the art or prepared as described below. These examples are being provided to further illustrate the present invention. They are for illustrative purposes only; the scope of the invention is not to be considered limited in any way thereby.

Unless otherwise stated, the following abbreviations have the stated meanings in the Examples below:

Me=methyl; Et=ethyl; Bu=butyl; Pr=propyl; Ph=phenyl; t-BOC=tert-butyloxycarbonyl; CBZ=carbobenzyloxy; and Ac=acetyl

10 DCC= dicyclohexylcarbodiimide

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DMAP=4-dimethylaminopyridine

DMF=dimethylformamide

EDCI= 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide

ESMS=Electron spray mass spectroscopy

15 FAB=Fast atom bombardment mass spectroscopy

HATU=O-(7-Azabenzotriazol-1-yl)-N,N,N',N'-tetramethyl uronium hexafluorophosphate

**HOBT= 1-hydroxybenzotriazole** 

LAH= lithium aluminum hydride

LDA= lithium diisopropylamide

20 NaBH(OAc)<sub>3</sub>= sodium triacetoxyborohydride

NBS=N-bromosuccinimide

PPA= polyphosphoric acid

RT=room temperature

TBAF=tetrabutylammonium fluoride

25 TBDMS=t-butyldimethylsilyl

TMEDA=N,N,N',N'-tetramethylethylenediamine

TEMPO=2,2,6,6-tetramethyl-1-piperidinyloxy, free radical

TLC=thin layer chromatography

**HRMS= High Resolution Mass Spectrometry** 

30 LRMS= Low Resolution Mass Spectrometry

nM= nanomolar

Ki= Dissociation Constant for substrate/receptor complex

pA2= -logEC<sub>50</sub>, as defined by J. Hey, Eur. J. Pharmacol., (1995), Vol. 294, 329-335.

Ci/mmol= Curie/mmol (a measure of specific activity)

<u>Step 1:</u>

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To a solution of 2-amino-4-methylpyridine (10.81 g, 100 mmol) in tert-butanol (250 ml) was added t-BOC anhydride (26.19 g, 120 mmol). The reaction mixture was stirred at 23 °C overnight, and then concentrated to an oil. The crude product was dry loaded onto a silica gel column and flash chromatographed (eluant: 30% hexanes-CH<sub>2</sub>Cl<sub>2</sub> to 0-2% acetone-CH<sub>2</sub>Cl<sub>2</sub>) to produce 15.25 g (73.32 mmol; 73%) of the desired product as a white solid.

<u>Step 2</u>:

N NHBOC-t

To a solution of the product of Step 1 (35.96 g, 173 mmol) in THF (1.4 l) at -78 °C was added a n-BuLi solution (1.4 M, 272 ml, 381 mmol) in hexanes portionwise over 30 min. The reaction mixture was then allowed to warm slowly and was stirred for 2 h at 23 °C, which resulted in the formation of an orange precipitate. The mixture was then cooled back to -78 °C, and pre-dried oxygen (passed through a Drierite column) was bubbled through the suspension for 6 h while the temperature was maintained at -78 °C. The color of the reaction mixture changed from orange to yellow during this time. The reaction was quenched at -78 °C with (CH<sub>3</sub>)<sub>2</sub>S (51.4 ml, 700 mmol) followed by AcOH (22 ml, 384 mmol) and allowed to warm with stirring to 23 °C. After 48 h, water was added and the product extracted into EtOAc. Purification by silica gel flash chromatography (eluant: 0–15% acetone/ CH<sub>2</sub>Cl<sub>2</sub>) provided 20.15 g (90 mmol; 52%) of the alcohol as a pale yellow solid.

25 Step 3:

To a solution of the product of Step 2 (19.15 g, 85.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (640 ml) was added a saturated aqueous solution of NaHCO<sub>3</sub> (8.62 g, 103 mmol) and

NaBr (444 mg, 4.3 mmol). The reaction mixture was cooled to 0 °C, and TEMPO (140 mg, 0.90 mmol) was introduced. Upon vigorous stirring, commercial bleach solution (122 ml, 0.7 M, 85.4 mmol) (5.25% in NaOCl) was added portionwise over 40 min. After an additional 20 min at 0 °C, the reaction mixture was quenched with saturated aqueous Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and allowed to warm to 23 °C. Dilution with water and extraction with CH<sub>2</sub>Cl<sub>2</sub>, followed by concentration and flash chromatography (eluant: 30% hexanes-CH<sub>2</sub>Cl<sub>2</sub> to 0-2% acetone-CH<sub>2</sub>Cl<sub>2</sub>) afforded 15.97 g (71.9 mmol; 84% yield) of the aldehyde as an off-white solid.

#### Step 4:

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To a solution of the product of Step 3 (11.87 g, 53.5 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (370 ml) was added ethyl isonipecotate (9.07 ml, 58.8 mmol) followed by four drops of AcOH. The reaction mixture was then stirred for 40 min at 23 °C, after which NaB(OAc)<sub>3</sub>H (22.68 g, 107 mmol) was added. The reaction mixture was stirred overnight at 23 °C, neutralized with saturated aqueous NaHCO<sub>3</sub>, diluted with water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. Concentration of the organic extracts, followed by silica gel flash chromatography (eluant: 0–4% sat. NH<sub>3</sub> in CH<sub>3</sub>OH-CH<sub>2</sub>Cl<sub>2</sub>) provided 19.09 g (52.6 mmol; 98%) of the ester as an off-white solid. Step 5:

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To a solution of the product of Step 4 (1.57 g, 4.33 mmol) in THF-water-CH<sub>3</sub>OH (10 ml of a 3:1:1 mixture) was added LiOH monohydrate (0.125 g, 5.21 mmol). The reaction mixture was stirred overnight at 23 °C, concentrated and exposed to high vacuum to obtain 1.59 g of crude title compound as a yellowish solid which was used without purification.

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#### Preparation 2

A solution of diamine 1B (see Method A, Step 1) (20g, 71.1mmol) and Et<sub>3</sub>N (30 ml, 213 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (400 ml) was cooled to 0 °C in an ice-water bath. To the well-stirred solution was added triphosgene (14.2 g, 47.3 mmol) cautiously (exotherm!) and portionwise over a period of 30 min. When addition was complete, stirring was continued at 0 °C for 1 h, then at RT for 16 h. The mixture was washed with 0.5N NaOH (200 ml), the organic layer was dried over anhydrous MgSO4 and concentrated under vacuum. Hot EtOAc (200 ml) was added to the semi-solid residue, and the resultant mixture was cooled to RT. Filtration yielded compound P2-1 as a white solid (16.5g); and silica gel flash chromatography [CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH (2N NH<sub>3</sub>) = 40:1] of the filtrate provided additional product as a white solid (2.7 g) [combined yield: 88%]. FABMS: 308 (MH+; 100%). Step 2:

P2-1 
$$\xrightarrow{POCl_3, \Delta}$$
  $\xrightarrow{Cl}$   $\xrightarrow{NC(O)OC_2H_5}$   $\xrightarrow{Cl}$   $\xrightarrow{NC(O)Cl}$   $\xrightarrow{NN}$   $\xrightarrow{NN}$   $\xrightarrow{P2-2}$   $\xrightarrow{F}$   $\xrightarrow{P2-10}$ 

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POCl<sub>3</sub> (100 ml) was added to **P2-1** (17.2 g; 56 mmol) in a round-bottomed flask flushed with dry N<sub>2</sub>. The mixture was placed in an oil bath heated to 108 °C and was maintained at reflux for 6 h. POCl<sub>3</sub> was then removed in vacuo. The residue was adjusted to pH ~ 9-10 with 7N methanolic ammonia and was concentrated to dryness under vacuum. CH<sub>2</sub>Cl<sub>2</sub> was added to the residue, insoluble material was filtered off, and the filtrate was again concentrated in vacuo. The residue was crystallized from EtOH to obtain compound **P2-2** as a white solid (12.6 g; 67%). ES-MS: 326.1 (MH<sup>+</sup>; 100%).

Varying amounts of compound P2-10 may be formed in this process and can be converted to desired product P2-2 by careful in situ treatment in CH<sub>2</sub>Cl<sub>2</sub> solution at 0 °C with one equivalent each of EtOH and NaH, followed by workup with ice-water and CH<sub>2</sub>Cl<sub>2</sub>. Low temperature is maintained in order to minimize reaction at the 2position of the benzimidazole nucleus.

Step 3:

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Sodium thiomethoxide (1.05 g; 15.0 mmol) was added to DMF (15 ml) in a round-bottomed flask flushed with N<sub>2</sub>. After stirring at RT for 30 min, solid chloride **P2-2** (3.25 g, 10 mmol) was added, and the resultant mixture was kept stirring at RT for 16 h. EtOAc (100 ml) and water (50 ml) were added to the reaction mixture. The aqueous layer was separated and further extracted with EtOAc (50 ml). The combined extracts were dried over anhydrous MgSO<sub>4</sub> and concentrated under vacuum. The residue was purified via flash chromatography on silica gel, eluting with EtOAc-hexanes (3:4), to obtain compound **P2-3** as a white solid (2.12 g; 63%). FABMS: 338.3 (MH<sup>+</sup>; 100%).

#### Step 4:

To a stirred solution of **P2-3** (300 mg, 12.5 mmol) in EtOH (40 ml)-isopropyl alcohol (40 ml) was added 25% (w/w) aqueous NaOH solution (20 ml). The resultant mixture was stirred at 85 °C for 24 h, then at 100 °C for an additional 4 h. Alcohols were removed under vacuum, and the aqueous residue was extracted sequentially with CH<sub>2</sub>Cl<sub>2</sub> (2 x 40 ml), then EtOAc (30 ml). Combined extracts were dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum. The residue was purified by silica gel flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>/2N methanolic ammonia = 12:1) to obtain Preparation 2 as an off-white solid (2.85 g, 70%). ES-MS: 266 (MH<sup>+</sup>; 100%).

#### **Preparation 3**

Step 1:

NaH (60 mg of a 60% dispersion; 1.48 mmol) was added to CH<sub>3</sub>OH (4 ml) in a flask charged with N<sub>2</sub>. After stirring at RT for 30 min, chloride **P2-2** (400 mg, 1.23 mmol) was added, and the resultant mixture was stirred at RT for 16 h. CH<sub>3</sub>OH was removed in vacuo, and to the residue were added CH<sub>2</sub>Cl<sub>2</sub> (30 ml) and water (10 ml). The organic layer was dried over anhydrous MgSO<sub>4</sub>, filtered, and the filtrate concentrated under vacuum. The residue was purified via flash chromatography on silica gel, eluting with EtOAc-hexanes (3:2) to obtain **P3-1** as a white foam (0.232g; 59%). ES-MS: 322.1 (MH<sup>+</sup>; 100%).

Step 2:

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1N aqueous KOH (4.82 mL; 4.82 mmol) was added to a solution of **P3-1** in EtOH (15 ml), and the resultant mixture was stirred at 80 °C for 48 h. The mixture was concentrated under vacuum. Water (3 ml) and CH<sub>2</sub>Cl<sub>2</sub> (15 ml) were added to the residue, and the organic layer was separated and dried over anhydrous MgSO<sub>4</sub>. Drying agent was filtered, and the filtrate was concentrated in vacuo to obtain Preparation 3 as a colorless glass (160mg; 95%). FABMS: 250.2 (MH<sup>+</sup>; 100%).

#### **Preparation 4**

P2-2 (300 mg; 0.923 mmol) and morpholine (3 ml) were mixed in a round-bottomed flask under N<sub>2</sub>, and the resultant mixture was heated to 80 °C for 16 h. Morpholine was removed under vacuum, and the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (20 ml). An insoluble white precipitate was filtered off, and the filtrate was concentrated and purified by means of flash chromatography on silica gel, eluting with CH<sub>2</sub>Cl<sub>2</sub>/2N methanolic ammonia (45:1), to obtain P4-1 as a colorless glass (0.325g; 94%). ES-MS: 377.1 (MH<sup>+</sup>; 100%).

#### Step 2:

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Trimethylsilyl iodide (240 microliters; 1.64mmol) was added to a solution of P4-1 (316 mg; 0.843 mmol) in CHCl<sub>3</sub> (2 ml) under N<sub>2</sub>, and the resultant solution was stirred at 55 °C for 7 h. The reaction was quenched with EtOH (2 ml), and the mixture was concentrated to dryness under vacuum. The residue was basified with a 1:1 (v/v) mixture of concentrated NH<sub>4</sub>OH and water to pH ~10 and extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 5 ml). The combined extracts were dried over anhydrous MgSO<sub>4</sub>. Drying agent was filtered, and the filtrate was concentrated under vacuum. The residue was purified via flash chromatography on silica gel, eluting with CH<sub>2</sub>Cl<sub>2</sub>-2N methanolic ammonia (13:1), to obtain compound Preparation 4 as a colorless glass. (181 mg; 70%). ES-MS: 305.1 (MH<sup>+</sup>; 100%).

Step 1:

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A solution of **P5-1** (3.5 g, 21 mmol) and **P5-2** (6.5 g, 38 mmol) in  $CH_2CI_2$  (3 ml) was heated to 110° C for 24 h and RT for 24 h. The reaction was diluted with  $CH_2CI_2$ , washed with water and brine, and dried ( $Na_2SO_4$ ). Purification on a flash column ( $SiO_2$ , 40% to 60% EtOAc in hexanes) gave **P5-3** (1.3 g, 21%; M+H = 295).

20 Step 2:

To a solution of **P5-3** (1.3 g, 4.4 mmol) in  $CH_3OH$  (30 ml) was added Ra-Ni (0.5 g) and the mixture was hydrogenated under a  $H_2$  atmosphere (50 psi) for 18 h.

Filtration through a pad of celite gave **P5-4** as a grey solid that was used without further purification (1.05 g, 90%; M+H = 265).

#### Step 3:

A solution of **P5-4** (1.05 g, 3.97 mmol), **P5-5** (0.49 g, 3.97 mmol), DEC (1.14 g, 5.96 mmol) and HOBT (0.8 g, 5.96 mmol) in  $CH_2Cl_2$  (10 ml) were stirred for 18 h at RT. The crude reaction mixture was diluted with additional  $CH_2Cl_2$  and washed with 5% aqueous NaOH and brine and dried (Na<sub>2</sub>SO<sub>4</sub>). Purification using flash chromatography (SiO, 8% EtOAc in hexane to 10%  $CH_3OH$  in EtOAc) gave **P5-6** (0.35 g, 24%; M+H = 370).

## <u>Step 4</u>:

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Compound **P5-6** (0.7 g, 1.89 mmol) was dissolved in HOAc (10 ml) and heated to 120° C for 3.5 h. The reaction was cooled to RT, concentrated in vacuo, neutralized by the addition of 10% aqueous NaOH and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated to give **P5-7** (0.58 g, 87%; M+H = 352) which was used in the next step without further purification.

Step 5:

A solution of **P5-7** (0.58 g, 1.65 mmol) and NaOH (0.43 g, 13.2 mmol) in EtOH/H<sub>2</sub>O (9/1, 10 ml) was heated to 100° C for 18 h. The reaction was cooled and concentrated and the residue purified on a flash column (SiO<sub>2</sub>, 10% CH<sub>3</sub>OH saturated with ammonia in CH<sub>2</sub>Cl<sub>2</sub>) to give Preparation 5 (0.42 g, 91%; M+H = 280).

#### Preparation 6

#### Step 1:

A solution of compound **P6-1** (prepared by procedures analogous to P2-1) (10.5 g, 36.2 mmol) and 2,6-di-*tert*-butylpyridine (12.2 ml, 54.4 mmol) in  $CH_2Cl_2$  (400 ml) was treated with 1M sol. of  $Et_3O^+BF_4^-$  (in  $CH_2Cl_2$ , 55 ml, 55 mmol). The reaction mixture was stirred at RT for 2h, quenched with 1N NaOH (100 ml), extracted with  $CH_2Cl_2$  (3x), dried with  $Na_2SO_4$  and concentrated. Purification by silica gel chromatography (eluant: 5-10% acetone/  $CH_2Cl_2$ ) to give 6.37 g of **P6-2** (20.0 mmol, 55%).

#### Step 2:

In a manner similar to that described in Preparation 3, Step 2, **P6-2** was converted to Preparation 6.

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#### Preparation 7

#### Step 1:

A mixture of **P7-1** (40g, 150 mmol), trimethyl orthoformate (66 ml, 64.0 g, 600 mmol) and a catalytic amount of *p*-toluenesulfonic acid monohydrate (300 mg, 1.58 mmol) was stirred under N<sub>2</sub> at 120 °C for 3 h. Excess orthoformate was removed under vacuum. The residue was partitioned between EtOAc (200 ml) and 1N NaOH (100 ml). The organic layer was washed with brine (100 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated

under vacuum. The residue was purified by silica gel flash chromatography  $(CH_2Cl_2/CH_3OH (2N NH_3) = 45:1)$  to obtain **P7-2** as a dark purple syrup (27.2 g, 66%), which solidified upon standing. ES-MS: 275 (MH $^+$ ; 100%).

Step 2:

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NBS was added portionwise (exotherm) to a solution of **P7-2** (27 g, 100 mmol) in CHCl<sub>3</sub> (300 ml), and the resultant solution was stirred at 60 °C for 16 h. Solvent was then removed under vacuum, and the residue was partitioned between EtOAc (200 ml) and 0.7N Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub> (250 ml). The organic layer was washed with brine (150 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum. The residue was purified by silica gel flash chromatography [CH<sub>2</sub>Cl<sub>2</sub>/acetone = 45:1] to obtain **P7-3** as a yellow solid (24.2 g, 69%). ES-MS: 353 (MH<sup>+</sup>; 100%). Step 3:

NaH (544 mg of a 60% dispersion, 13.6 mmol) was added to a solution of CH<sub>3</sub>OH (0.551 ml, 436 mg, 13.6 mmol) in DMF (5 ml). The resultant mixture was stirred at RT for 30 min before adding solid bromide **P7-3** (3.99 g, 11.3 mmol). The reaction suspension was stirred at RT for 16 h. The mixture was then partitioned between EtOAc (800 ml) and water (40 ml). The aqueous layer was extracted with EtOAc (40 ml). Combined extracts were washed with brine (30 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum to obtain Preparation 7 as a white syrup (2.81 g, 81%), which was used without further purification. ES-MS: 305 (MH<sup>+</sup>; 100%).

#### Preparation 8

A solution of **1B** (15 g, 52.8 mmol) and 1,1'-thiocarbonyldiimidazole (25 g, 140 mmol) in THF (300 ml) was stirred at 72  $^{\circ}$ C under N<sub>2</sub> for 16 h, during which time a precipitate formed. THF was removed under vacuum, and the residue was purified by silica gel flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>/acetone = 20:1) to obtain **P8-1** as a light yellow solid (16.7 g, >95%). ES-MS: 324 (MH<sup>+</sup>; 100%).

#### Step 2:

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To a stirred mixture of **P8-1** (4.00 g, 12.5 mmol) and K<sub>2</sub>CO<sub>3</sub> (2.05 g, 13.6 mmol) in DMF (40 ml) under a N<sub>2</sub> atmosphere was added CH<sub>3</sub>I (0.85 ml, 1.94 g, 13.6 mmol). The resultant mixture was stirred at RT for 16 h before partitioning between EtOAc (100 ml) and water (40 ml). The aqueous layer was extracted with EtOAc (40 ml). Combined extracts were washed with brine (30 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum to obtain Preparation 8 as a foamy white solid (4.20 g, >95%; contained a small amount of DMF), which was used without further purification. ES-MS: 338 (MH<sup>+</sup>; 100%).

#### Prenaration 0

#### 20 Step 1:

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(Modified published procedure: G. Heinisch, E. Luszczak, and M. Pailer: *Monatshefte für Chemie*, 1973 (104), 1372.

**P9-1** (4.5 g, 47.8 mmoles), **P9-2** (8.12g, 76.5 mmoles), and anhydrous ZnCl<sub>2</sub> were heated, under N<sub>2</sub>, in a dry apparatus, at a bath temperature of 160 °C for 5 h.

The resulting oil was purified by flash chromatography on silica gel using 30% Hexanes/EtOAc, yielding 5.92 grams (67%) of the product.

Step 2:

OsO<sub>4</sub> (5.0 ml in t-butanol, 2.5% w/w) was added to **P9-3** (5.9 g, 32.38 mmoles) dissolved in p-dioxane (87 ml) and water (29 ml). NalO<sub>4</sub> (14.1 g, 65.92 mmoles) was added, with good stirring, in small portions, over a period of 6 h. The mixture was then diluted with p-dioxane and filtered. After removing most of the solvent under reduced pressure, the residue was taken in CH<sub>2</sub>Cl<sub>2</sub> (600 ml) and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After removal of the solvent, the mixture was purified by flash chromatography on silica gel using 5% CH<sub>3</sub>OH/CH<sub>2</sub>Cl<sub>2</sub> as eluent to obtain Preparation 9. Yield: 2.89 g (82%).

#### **Preparation 10**

Step 1:

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A solution of **P10-1** (2 g, 15 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 ml) was treated with Et<sub>3</sub>N (3 g, 30 mmol) and triphenylmethyl chloride (TrCl, 4.25 g, 15.3 mmol) and stirred at RT overnight. The solvent was removed in vacuo and the residue purified via flash column chromatography (SiO<sub>2</sub>, 20% EtOAc in hexane) to give **P10-2** (5.2 g, 46%). Step 2:

A solution of **P10-2** (5.2 g, 14.6 mmol) in CCl<sub>4</sub> (80 ml) was treated with NBS (7.8 g, 43 mmol) and the reaction heated to 80° C overnight. The reaction was cooled, filtered and concentrated, and the residue was purified via flash column chromatography (SiO<sub>2</sub>, 20% to 30% EtOAc in hexane) to give Preparation 10 (2.8 g, 42%, M+H = 453, 455)

Step 1:

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To a stirred solution of **P8-1** (6.5 g, 20.1 mmol) in EtOH (80 ml) was added 25% (w/w) aqueous NaOH solution (20 ml). The resultant mixture was stirred at 90 °C for 16 h. EtOH was removed under vacuum, and the residue was adsorbed directly onto silica gel and subjected to flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>/2N methanolic ammonia = 9:1) to obtain **P11-1** as a white solid (4.46 g, 70%). ES-MS: 252 (MH<sup>+</sup>; 100%).

Step 2:

A mixture of P11-1 (3.95 g; 15.7 mmol), BOC-isonipecotic acid (3.60 g; 15.7 mmol), HOBT (3.19 g; 23.6 mmol), DIPEA (3ml; 2.23g; 17.2 mmol) and EDCI (4.50 g; 23.6 mmol) in DMF (30 ml) was stirred under  $N_2$  at RT for 16 h. The reaction mixture was partitioned between EtOAc (60 ml) and water (40 ml). The aqueous phase was extracted with EtOAc (40 ml), and the combined extracts were washed with brine (40 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum. The residue was purified by silica gel flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH (2N NH<sub>3</sub>) = 40:1) to obtain P11-2 as a white solid (~7.3 g, ~100%), containing a small amount of DMF, used without further purification in Step 3 below. ES-MS: 463 (MH<sup>+</sup>; 70%); 407 (100%). Step 3:

To a stirred mixture of P11-2 (460 mg; 1 mmol) and K<sub>2</sub>CO<sub>3</sub> (165 mg; 1.20 mmol) in DMF (4 ml) under a N<sub>2</sub> atmosphere was added Etl (92 microliters; 179 mg; 1.15 mmol). The resultant mixture was stirred at RT for 16 h and was then partitioned between EtOAc (20 ml) and water (10 ml). The aqueous phase was extracted with EtOAc (10 ml), and the combined extracts were washed with brine (20 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum to obtain P11-3 as a pale yellow foam (471 mg, 96%), containing a small amount of DMF, used without further purification in Step 4 below. ES-MS: 463 (MH<sup>+</sup>; 85%); 435 (100%).

#### 10 <u>Step 4</u>:

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To a solution of **P11-3** (465 mg; 0.949 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (4 ml) was added TFA (1 ml; 1.54 g; 13.5 mmol). The resultant solution was stirred for 2 h at RT and was then partitioned between CH<sub>2</sub>Cl<sub>2</sub> (20 ml) and 1:1 (v/v) concentrated NH<sub>4</sub>OH:water (5 ml). The aqueous phase was extracted successively with 95:5 CH<sub>2</sub>CL<sub>2</sub>:EtOH (5 ml) and EtOAc (5 ml). The combined extracts were dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum to obtain Preparation 11 as a pale white foam (353 mg, 95%), used without further purification. ES-MS: 391 (MH<sup>+</sup>; 100%).

### Example 1

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$$\frac{\text{Method A}}{\text{Step 1:}}$$

$$\downarrow NO_2 \qquad H_2N \qquad \qquad \downarrow NO_2 \qquad NO_2 \qquad$$

A mixture of **a** (25 g, 0.16 mol), **b** (27 g, 0.16 mol),  $K_2CO_3$  (26 g, 0.19 mol), and NaI (2.4g, 0.016 mol) in dimethylacetamide (50 ml) was heated at 140 °C for 3.5 h.

The reaction mixture was concentrated to one-third volume, poured onto saturated aqueous NaHCO<sub>3</sub>, and extracted with EtOAc (4×). The combined organic layers were washed with water (2×) and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Recrystallization with EtOH provided **1A** (48 g, 98%).

A suspension of **1A** (20.00 g, 64.2 mmol,) and Raney® 2800 Nickel (5.0 g) in ethanol (70 ml) and THF (140 ml) was shaken under  $H_2$  (40 psi) for 2 h. The mixture was filtered through a short pad of celite. The filtrate was concentrated and dried on

#### Step 2:

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vacuum to deliver a tan solid (18.20 g, ~100%).

A solution of **1B** (5.00 g, 17.77 mmol) and picolinoyl chloride hydrochloride (3.16g, 17.75 mmol) in  $CH_2Cl_2$  (400 ml) and  $Et_3N$  (15 ml) was stirred at RT. After 15 h, the reaction was diluted with  $CH_2Cl_2$ , washed with water, dried over  $Na_2SO_4$ , concentrated, and dried on vacuum to provide a brown foam (6.47g, 94%).

#### Step 3:

$$1C \longrightarrow \bigvee_{N=0}^{N} \bigvee_{N=0}^{N}$$

A solution of 1C (1.77g, 4.58 mmol) in ethanol (50 ml) and concentrated H<sub>2</sub>SO<sub>4</sub> (5.0 ml) was refluxed for 3 h, cooled to RT, and neutralized with 1.0 M NaOH until pH = 10. The resulting mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated on reduced pressure. The residue was purified by flash chromatography (silica gel, 5% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub> as eluent) to provide a tan foam (1.58g, 94%).

#### Step 4:

lodotrimethylsilane (6.30g, 31.48 mmol) was added to a solution of 1D (3.88g, 10.53 mmol) in anhydrous 1,2-dichloroethane (40 ml). The resulting solution was stirred at 75 °C for 4 hours, cooled to RT, and treated with 1.0 M NaOH solution. The mixture was then extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined extracts were washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and the solvent evaporated. Purification of the residue by flash chromatography (silica gel, 10% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub> as eluent) delivered an off-white foam (2.10g, 67%).

#### Step 5:

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Amine 1E (5.80g, 19.6 mmol) and Preparation 1 (5.32g, 23.4 mmol) were dissolved in DMF (60 ml) and CH<sub>2</sub>Cl<sub>2</sub> (60 ml). To the resulting solution, EDCI hydrochloride (5.70g, 24.50 mmol), HOBT (1.30g, 24.50 mmol), and diisopropylethylamine (5.08g, 39.6 mmol) were added successively. The resulting reaction mixture was stirred at 70°C for 4 hours, cooled to RT, diluted with CH<sub>2</sub>Cl<sub>2</sub>, washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Flash chromatography (SiO<sub>2</sub>, 5% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub>  $\rightarrow$  90:10:0.5 CH<sub>2</sub>Cl<sub>2</sub>:CH<sub>3</sub>OH:NH<sub>4</sub>OH) of the residue provided a tan foam (7.89g, 65%).

A solution of **1F** (7.89g, 12.88 mmol) and TFA (29g, 257 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (65 ml) was stirred at RT for 12 h, neutralized with 1.0 M NaOH, and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were washed with water, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. Purification of the crude product by flash chromatography (SiO<sub>2</sub>, 5% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub> to 90:10:0.5 CH<sub>2</sub>Cl<sub>2</sub>:CH<sub>3</sub>OH:NH<sub>4</sub>OH) provided the title compound as a white solid (5.80g, 88%). MS: 514 (MH<sup>+</sup>).

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#### Method B

#### Step 1:

TFA (200 ml, 2.596 mol) was added to a solution of **2A** (20g, 51.36 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (100 ml). The resulting reaction mixture was stirred at RT for 6 h, neutralized with 1.0 M NaOH, and extracted. The combined extracts were washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Flash chromatography gave an orange solid (13.50g, 91%).

#### Step 2:

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Amine **2B** (1.50g, 5.19 mmol) and Preparation 1 (1.75g, 5.13 mmol) were dissolved in DMF (10 ml) and CH<sub>2</sub>Cl<sub>2</sub> (10 ml). To the resulting solution, EDCl hydrochloride (1.50g, 7.83 mmol), HOBT (1.05g, 7.82 mmol), and diisopropylethylamine (3.71g, 28.70 mmol) were added successively. The resulting reaction mixture was stirred at 70°C for 18 h, cooled to RT, diluted with CH<sub>2</sub>Cl<sub>2</sub>, washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Flash chromatography of the residue provided an orange gel (2.31g, 74%).

#### Step 3:

A suspension of **2C** (2.10 g, 3.46 mmol,) and Raney® 2800 Nickel (1.0 g) in CH<sub>3</sub>OH (100 ml) was shaken under H<sub>2</sub> (30 psi) for 6 h. The mixture was filtered through a short pad packed with celite. The filtrate was concentrated and dried on vacuum to deliver an orange solid (1.80 g, 90%).

#### Step 4:

Amine **2D** (200 mg, 0.347 mmol) and picolinoyl chloride hydrochloride (62 mg, 0.348 mmol) were dissolved in CH<sub>2</sub>Cl<sub>2</sub>. Et<sub>3</sub>N was then introduced via a syringe. The resulting solution was stirred at RT for 6 h, treated with 1.0 M NaOH solution, and extracted. The extracts were washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Purification of the residue by flash chromatography gave a white foam (167 mg, 71% yield).

#### Step\_5:

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A solution of **2E** (160 mg, 0.235 mmol) and H<sub>2</sub>SO<sub>4</sub> (concentrated, 0.50 ml) in ethanol (10 ml) was refluxed for 2.5 h, cooled to RT, and neutralized with 1.0 M NaOH. After extraction of the mixture, the combined organic layers were washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Purification of the crude product using prep TLC (10% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub>) provided the title compound as a white solid (88 mg, 66%). MS: 564 (MH<sup>+</sup>)

15 Method D

Step 1:

$$CI$$
 $NH_2$ 
 $NH$ 

Diamine **3A** (1.43 g, 10 mmol) and isonipecotic acid **3B** (1.29 g, 10 mmol) were mixed, and PPA (20 g) was added. The resulting mixture was heated at 180 °C for 3.5 h, cooled to RT and diluted with water to 100 ml. The solution was then basified with solid NaOH to pH 14. The resultant copious precipitate was filtered off. The precipitate was washed repeatedly with CH<sub>3</sub>OH, and combined CH<sub>3</sub>OH extracts were concentrated-dry loaded on silica gel and flash chromatographed (25-40% 5N NH<sub>3</sub> in CH<sub>3</sub>OH/ CH<sub>2</sub>Cl<sub>2</sub>) to provide **3C** as a dark solid (1.90 g, 81%).

Step 2:

To the mixture of acid **3D** (181mg, 0.54 mmol), HATU (247 mg, 0.65 mmol) and  $Et_3N$  (84  $\mu$ l, 0.6 mmol) in DMF (12 ml) was added amine **3C** (126 mg, 0.54 mmol). The resulting mixture was stirred at RT for 24 h, concentrated, redissolved in CH<sub>3</sub>OH, concentrated-dry loaded on silica gel and flash chromatographed (5-10% 5N NH<sub>3</sub> in CH<sub>3</sub>OH/ CH<sub>2</sub>Cl<sub>2</sub>) to provide **3E** as a yellow oil (210mg, 70%). Step 3:

A solution of **3E** (96 mg, 0.174 mmol) in 15 ml of 1M HCl in 25% CH<sub>3</sub>OH/dioxane was stirred at RT for 48 h. The mixture was concentrated, exposed to high vacuum, redissolved in CH<sub>3</sub>OH, concentrated-dry loaded on silica gel and flash chromatographed (10-15% 5N NH<sub>3</sub> in CH<sub>3</sub>OH/ CH<sub>2</sub>Cl<sub>2</sub>) to provide the title compound as a colorless oil (48 mg, 61%). MS: 453 (MH<sup>+</sup>)

#### Method E

## 15 Step 1:

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A mixture of neat **4A** (1.75g, 6.66 mmol) and **4B** (2.93g, 15.07 mmol) was stirred at 120 °C for 2 days, cooled to RT, treated with 1.0 M NaOH solution (30 ml), and extracted with EtOAc. The combined organic layers were washed with water and dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation to dryness, the crude residue was flash chromatographed (silica gel, 50% EtOAc in hexanes as eluent) to give 510 mg of **4C** (18%).

#### Step 2:

$$4C \longrightarrow \bigvee_{N} \stackrel{H}{\bigvee_{N}} \bigvee_{CF_{3}} \stackrel{NH}{\downarrow}$$

To a 500 ml pressure bottle was added **4C** (490 mg, 1.18 mmol) in CH<sub>3</sub>OH (20 ml). Under N<sub>2</sub> stream, palladium hydroxide (300 mg, 20 wt.% on carbon) solid was added. The reaction mixture was shaken under 55 psi of hydrogen for 40 h and

filtered. The filtrate was concentrated and dried on vacuum to deliver a yellow solid (340 mg, 88%).

## Step 3:

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To a 50 ml round-bottomed flask were successively added **4D** (287 mg, 0.88 mmol), Preparation 1 (300 mg, 0.88 mmol), EDCI hydrochloride (210 mg, 1.10 mmol), HOBT (149 mg, 1.10 mmol), and diisopropylethylamine (228 mg, 1.76 mmol). DMF (3 ml) and CH<sub>2</sub>Cl<sub>2</sub> (3 ml) were introduced via a syringe. The resulting reaction mixture was stirred at 70 °C for 15 h and cooled to RT. After addition of 1 N NaHCO<sub>3</sub> solution, the resulting mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic solutions were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. Purification of the crude product by flash chromatography on silica gel with 10% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub> as the eluent provided **4E** as a solid (231 mg, 41%).

## Step 4:

To a 25 ml round-bottomed flask was added 4E (200 mg, 0.31 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (2.5 ml). TFA was then introduced via a syringe. The resulting solution was stirred at RT for 15 h, diluted with CH<sub>2</sub>Cl<sub>2</sub>, neutralized with 1.0 M NaOH solution, and separated. The organic solution was washed with water and dried over Na<sub>2</sub>SO<sub>4</sub>. After evaporation of the solvent, the crude product was purified on a preparative TLC plate with 10% CH<sub>3</sub>OH in CH<sub>2</sub>Cl<sub>2</sub> as the eluent to provide the title compound as a white solid (85 mg, 50%). MS: 544 (MH<sup>+</sup>).

Step 1:

A solution of compound **5A** (100g, 0.389 mol) in THF (400 ml) was added dropwise over 1.0 h to a solution of LDA (233 mL, 2.0 M in THF/heptane/ethyl-

benzene, 0.466 mol) in THF (300ml) at 0 °C. The red-orange solution was stirred at 0 °C for 30 min, and then transferred by cannula to a pre-cooled (0 °C) solution of N-fluorobenzenesulfonimide (153 g, 0.485 mol) in dry THF (600 ml). The reaction mixture was stirred at 0 °C for 30 min, and then at 20 °C for 18 h. The total solvent volume was reduced to approximately one third, and EtOAc (11) was added. The solution was washed successively with water, 0.1 N aq. HCl, saturated aq. NaHCO<sub>3</sub>, and brine. The organic layer was dried over MgSO<sub>4</sub>, filtered, and concentrated under reduced pressure to yield a crude liquid. Separation by flash chromatography (6:1 hexanes-EtOAc) gave compound **5B** (93.5 g, 87%).

## 10 <u>Step 2</u>:

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A solution of **5B** (50g, 0.181 mol) in THF (300 ml) and CH<sub>3</sub>OH (200 ml) was treated with a solution of LiOH-H<sub>2</sub>O (9.2 g, 0.218 mol) in water (100 ml) and then heated to 45 °C for 6 h. The mixture was then concentrated and dried in vacuo to provide **5C** (45 g, 100%).

#### Step 3:

Compound **5C** (20.4 g, 0.081 mol) was added slowly to a stirred flask of  $CH_2Cl_2$  (250 ml) at 20 °C. The resulting white slurry was cooled to 0 °C and treated slowly with oxally chloride (6.7 ml, 0.075 mol) and a drop of DMF. After stirring at 20 °C for 0.5 h, the mixture was concentrated and dried in vacuo to provide **5D**. Step 4A:

A mixture of **c** (64 g, 0.40 mol), **d** (84 ml, 0.52 mol), and K<sub>2</sub>CO<sub>3</sub> (66 g, 0.48 mol) in anhydrous toluene (350 ml) was heated at reflux overnight. The reaction mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub>, washed three times with 5% aqueous NaOH, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Recrystallization with MeOH provided **e** (121 g, ~100%) as a yellow solid.

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A suspension of e (121 g, 0.41 mol) and Raney Nickel (10 g) in EtOH (400 ml) was shaken under  $H_2$  (40 psi) for 4 h. The mixture was filtered through a short pad of Celite (washing with CH<sub>3</sub>OH). The filtrate was concentrated and dried in vacuo to provide f (109 g, ~100%) as a dark brown solid.

A solution of **f** (109 g, 0.41 mol) in CH<sub>2</sub>Cl<sub>2</sub>-DMF (1:1, 500 ml) was treated with picolinic acid (61 g, 0.50 mol), EDCI (119 g, 0.62 mol), HOBt (84 g, 0.62 mol) and iPr<sub>2</sub>NEt (141 ml, 1.03 mol). The mixture was stirred at 70 °C for 6 h and then overnight at 20 °C. The reaction mixture was diluted with EtOAc, washed 3 times with 5% aqueous NaOH, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Flash chromatography (0–100% EtOAc/hexane) provided **g** (131 g, 86%).

A solution of **g** (131 g, 0.36 mol) in AcOH (200 ml) was heated at 120 °C overnight. The reaction mixture was cooled, carefully basified with 5% aqueous NaOH and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. Flash chromatography (0–80% EtOAc/hexane) provided **h** (95 g, 76%) as a yellow solid.

A solution of **h** (95 g, 0.27 mol) in anhydrous CHCl<sub>3</sub> (300 ml) was treated with iodotrimethylsilane (272 g, 1.36 mol) and heated at 70 °C for 5 h. The reaction mixture was cooled, quenched with cold 10% aqueous NaOH, and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. Flash chromatography (2N NH<sub>3</sub>-CH<sub>3</sub>OH/EtOAc) provided **5E** (43 g, 57%) as a pale yellow solid.

#### Step 4B:

A mixture of **5D** (0.075 mol) in CH<sub>2</sub>Cl<sub>2</sub> (250 ml) was treated with **5E** (15 g, 0.054 mol) and iPr<sub>2</sub>NEt (25 ml, 0.135 mol) while maintaining a temperature of 20 °C. After 1 h, the mixture was concentrated and then stirred in CH<sub>3</sub>OH (200 ml)/CH<sub>2</sub>Cl<sub>2</sub> (200 ml)/H<sub>2</sub>O (1 ml) for 1 h at 20 °C. The solvent was then evaporated. Treatment with TFA (200 ml) in CH<sub>2</sub>Cl<sub>2</sub> (250 ml) at 20 °C followed by flash chromatography (0–7% 7N NH<sub>3</sub>-CH<sub>3</sub>OH/CH<sub>2</sub>Cl<sub>2</sub>) provided **5F** (80–90% from **5C**). Step **5**:

# 10 Method A:

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A solution of **5F** (0.41 g, 1.0 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) was treated with **5G** (0.31 g, 2.5 mmol, JP Patent 63227573, 1988), NaBH(OAc)<sub>3</sub> (0.53 g, 2.5 mmol) and few drops of AcOH and then stirred overnight at 20 °C. The mixture was partitioned between 10% NaOH and CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried with Na<sub>2</sub>SO<sub>4</sub> and concentrated. Flash chromatography (0–5% 7N NH<sub>3</sub>-CH<sub>3</sub>OH/CH<sub>2</sub>Cl<sub>2</sub>) provided the title compound (0.45g, 87%). MS: 516 (M+H).

#### Method B:

A solution of **5G** (50 g, 0.41 mol) in CH<sub>3</sub>OH (300 ml) was cooled to 0 °C and carefully treated with NaBH<sub>4</sub> (20g, 0.53 mol in 6 batches) over 20 min. The reaction was then allowed to warm to 20 °C and was stirred for 4 h. The mixture was again cooled to 0 °C, carefully quenched with saturated aqueous NH<sub>4</sub>Cl, and concentrated. Flash chromatography (5–10% 7N NH<sub>3</sub>-CH<sub>3</sub>OH/CH<sub>2</sub>Cl<sub>2</sub>) provided **5H** (31g, 62%) as a light yellow solid.

A slurry of **5H** (31 g, 0.25 mol) in CH<sub>2</sub>Cl<sub>2</sub> (500 ml) was cooled to 0 °C and slowly treated with SOCl<sub>2</sub> (55ml, 0.74 mol over 30 min). The reaction was then stirred overnight at 20 °C. The material was concentrated, slurried in acetone, and then filtered. The resulting beige solid **5I** was dried overnight in vacuo (38.4g, 52%, HCl salt).

A homogeneous solution of **5F** (16.4 g, 40 mmol) in anhydrous DMF (200 ml) was cooled to 0 °C, carefully treated with NaH (8g, 200 mmol), and stirred at 20 °C for 20 min. The reaction mixture was then cooled to 0 °C, treated with NaI (6g, 40 mmol) and **5I** (14.5g, 80 mmol), and then stirred overnight at 20 °C. The reaction was diluted with CH<sub>2</sub>CI<sub>2</sub> (500 ml), washed with 1N aqueous NaOH, washed with brine, filtered through Celite, and concentrated. Flash chromatography (0–4% 7N NH<sub>3</sub>-CH<sub>3</sub>OH/CH<sub>2</sub>CI<sub>2</sub>) provided Ex. 5 (16.9g, 82%) as a beige solid.

To a stirred solution of diamine **1B** (1.0g, 3.55 mmol) in C<sub>2</sub>H<sub>5</sub>OH (25 ml), at RT was added portionwise solid CNBr (564 mg; 5.33 mmol). The resultant solution was allowed to stir at RT for 5 days before removing solvent under vacuum. The residual oil was partitioned between EtOAc (30 ml) and 2M Na<sub>2</sub>CO<sub>3</sub> (10 ml). The aqueous layer was adjusted to pH ~10 by addition of a few drops of 6N NaOH and was then reextracted with EtOAc (2 x 10 ml). Combined extracts were washed with brine (5 ml) and filtered through anhydrous MgSO<sub>4</sub>. The filtrate was stripped in vacuo to obtain compound **6A** as brown powder (1.03 g; 94%) sufficiently pure for use without purification. FABMS: 307 (MH<sup>+</sup>; 100%).

Step 2:

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In a dry flask, under an inert atmosphere, a mixture of compound **6A** (369 mg; 1.20 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (11 ml) was stirred and sonicated until the formation of a

clear, amber solution to which was added via syringe 4-fluorophenyl isocyanate (158 microliters; 190 mg; 1.38 mmol). After 30.5 h at RT, a few drops of CH<sub>3</sub>OH were added to the reaction solution, and solvent was removed under vacuum. The residual solid was dissolved in boiling Et<sub>2</sub>O (~30 ml). Insoluble matter was filtered, and the filtrate was diluted to a volume of ~60 ml with hot hexanes. The solution was concentrated on a steam bath to a volume of ~30 ml, by which point precipitation had begun. The mixture was allowed to stand at RT for ~3 h. Filtration and washing with Et<sub>2</sub>O-hexanes (1:1 v/v) yielded compound **6B** as a reddish-brown powder (394 mg; 74%). FABMS: 444 (MH\*; 100%). Although TLC and NMR indicated the presence of minor impurities, the product was sufficiently pure for use in Step 3 below. Step 3:

To a stirred suspension of compound **6B** (333 mg; 0.751 mmol) in CHCl<sub>3</sub> (2 ml), contained in a flask equipped for reflux under an inert atmosphere, was added via syringe (CH<sub>3</sub>)<sub>3</sub>Sil (214 microliters; 301 mg; 1.51 mmol). Solids dissolved rapidly to produce a dark reddish-brown solution. Stirring was continued at RT for 20 min before placing the reaction mixture in an oil bath preheated to 50 °C. After 5 h at 50 °C, a second portion of (CH<sub>3</sub>)<sub>3</sub>Sil (54 microliters; 75 mg; 0.378 mmol) was added and heating continued at 50 °C for another 2.5 h. The reaction mixture (consisting of solid and solution phases) was removed from the heating bath and was treated with CH<sub>3</sub>OH (2.5 ml) added in two portions. The reaction mixture was stirred and warmed to 50 °C for a few minutes, allowed to cool and was then filtered. Collected solids were washed with 1:1 (v/v) CH<sub>3</sub>OH-EtOAc to obtain the hydriodide salt form of **6C** as a pale reddish-brown powder (356 mg) wich was used in the next step without further purification. FABMS: 372 (MH<sup>+</sup>; 100%).

Step 4:

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To a stirred suspension of **6C** (340 mg; 0.681 mmol), Prep. 1 (228 mg; 0.681 mmol), HOBT (9.2 mg; 0.0681 mmol) and NEt<sub>3</sub> (379 microliters; 275 mg; 2.72 mmol)

in DMF (13 ml) was added solid EDCI (163 mg; 0.851 mmol). The cloudy reaction mixture was placed in a preheated oil bath and was stirred at 50 °C for 30 min, after which the resultant clear, amber solution was stirred for 23.5 h at RT. A few drops of water were added, and the reaction mixture was concentrated at 60 °C under vacuum. The concentrate was partitioned between EtOAc (20 ml) and water (5 ml)-brine (2.5 ml). The aqueous phase was extracted with EtOAc (2 x 5 ml). Combined extracts were washed with brine (2.5 ml) and filtered through anhydrous MgSO<sub>4</sub>. The filtrate was evaporated under vacuum, and the residue was purified by flash chromatography on silica gel, eluting with a gradient of CH<sub>2</sub>Cl<sub>2</sub>-CH<sub>3</sub>OH-NH<sub>4</sub>OH (97:3:0.5 -> 96:4:0.5). Product 6D (222 mg; 47%) was obtained as pale yellow powder. FABMS: 689 (MH<sup>+</sup>; ~93%); 578 (~58%); 478 (100%).

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To a solution of **6D** (208 mg; 0.302 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (3 ml) was added TFA (928 microliters; 1.37 g; 12.1 mmol) with swirling of the flask, which was then flushed with dry N<sub>2</sub>, sealed and allowed to stand at RT for 6 h. The reaction solution was evaporated under vacuum, and the residue was partitioned between EtOAc (20 ml) and 2M Na<sub>2</sub>CO<sub>3</sub> (3 ml) plus sufficient water to produce two clear phases. The aqueous phase was extracted with EtOAc (3 x 5 ml). Combined extracts were washed with brine (3 ml) and filtered through anhydrous MgSO<sub>4</sub>. The filtrate was stripped of solvent in vacuo, and the residue was subjected to flash chromatography on silica gel, eluting with CH<sub>2</sub>Cl<sub>2</sub>-CH<sub>3</sub>OH-NH<sub>4</sub>OH (97:3:0.5). The title compound (130 mg; 72%) was obtained as pale yellow powder. FABMS: 589 (MH<sup>+</sup>; ~64%); 478 (100%).

Using procedures similar to those described above, employing the appropriate starting materials, compounds in the following tables are prepared:

No.	R	R <sup>25</sup>	R <sup>3</sup>	R <sup>13</sup>	Z	R <sup>6</sup>	Physical Data MS (MH*)
7	-CH₃	5-OCH <sub>3</sub>	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	463
8	-CH <sub>3</sub>	6-CI	Н	Н	-CH₂-	2-NH <sub>2</sub>	467
9	-CH <sub>3</sub>	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	467

10	-CH <sub>3</sub>	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	512
11		5-Cl	Н	H	-CH <sub>2</sub> -	2-NH <sub>2</sub>	535
	$\bigcirc$ *	J-01	,,		J. 12	Z-141 12	JJ55
12	benzyl	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	527
13	-CH(CH <sub>3</sub> ) <sub>2</sub>	5-Br	H	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	540
14	-CH <sub>2</sub> NH <sub>2</sub>	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	488
15	-CH₂NHSO₂CH₃	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	526
16	-CH <sub>2</sub> NHC(O)CH <sub>3</sub>	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	524
17	-CH <sub>2</sub> OCH <sub>3</sub>	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	481
18	-CH <sub>2</sub> NH <sub>2</sub>	5-CI	Н	H	-CH <sub>2</sub> -	2-NH <sub>2</sub>	482
19	-CH <sub>2</sub> OCH <sub>3</sub>	6,7-di-F	Н	H .	-CH₂-	2-NH <sub>2</sub>	499
20	<b>€</b>	6-F	Ĥ	Н	-CH₂-	2-NH <sub>2</sub>	521
21	$\bigcirc^{\sim}$	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	521
22	C)-5	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	507
23	0 N 3	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	520
24	0 N - 5	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	521
25	<b>(</b> }-}	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	568
26	$\stackrel{\sim}{\triangleright}$	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	507
27	( <del>)</del> -{}	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	507
28	F-{-}-{-}	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
29	F	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	549
30	F-{-}	6-F	Н	Н	-CH₂-	2-NH₂	531
31	F <del>−</del> √>-Şi F	6,7-di-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	567
32	F-{-}-};	6-Cl	Н	Н	-CH₂-	2-NH₂	547
33	F-(	5-F	Н	Н	-CH₂-	2-NH₂	531
34	F F	5-CI	H	Н	-CH₂-	2-NH <sub>2</sub>	565

			т	<del>,</del>	,		
35	F—————————————————————————————————————	Н	H	H	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
36	F-(-)	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	547
37		5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	529
38	ST.	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	557
39	<b>₹</b>	5-Br	Н	Н	-CH₂-	2-NH₂	592
40	F-\$	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH₂	610
41	cı—{	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	547
42	OH OH	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	529
43	(T)-{	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	553
44	Q <sub>N</sub> <sub>z</sub> z	6-F	Н	Н	-CH₂-	2-NH₂	564
45	cı—{_}	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	529
46	CI ~~~	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	581
47	cı—{_}	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	563
48	cı—{_}	6-CI	Н	Н	-CH <sub>2</sub> -	2-NH₂	563
49	€ξ. осн₃	5-F	Н	Н	-CH₂-	2-NH₂	543
50	<b>ℂ</b> }–ξ <sub>2</sub> CF <sub>3</sub>	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	581
51	CI ~~ }	5-Cl	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	597
52	CV−Ş; OCF <sub>3</sub>	5-F	Н	Н	-CH₂-	2-NH₂	597
53	Су-}-}	5-Br	Н	Н	-CH₂-	2-NH <sub>2</sub>	604

54	CI ~~~	6-CI	Н	Н	-CH₂-	2-NH <sub>2</sub>	597
55	H₃CH₂CO-	5-CH₃	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	571
56	F₃C F₃C	5-CI	Н	Н	-CH₂-	2-NH <sub>2</sub>	665
57	F <sub>3</sub> C F <sub>3</sub> C	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	710
58	<b>₹</b>	6-ethoxy	H	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	540
59	~\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	5-CI .	Н	Н	-CH₂-	2-NH <sub>2</sub>	546
60	√Z N−Z NH₂	Н	Н	Н	-CH <sub>2</sub> -	2-NH₂	511
61	<b>₹</b>	5-F	H	Н	-CH <sub>2</sub> -	Н	499
62	€ <u>N</u> -5	6-CI	Н	Н	-CH <sub>2</sub> -	2-NH₂	530
63	<b>n</b> _}-Ş:	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	515
64	<b>₹</b>	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	514
65	N->-5	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	515
66		7-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
67	₹ <u>N</u> -3	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	496
68	<u>₹</u> }-}	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	515
69	<u>√</u> <u>N</u> - ξ	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
70	ν <u>=</u> Ν_ξ	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
71	<b>€</b> N-5	5,6-di-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	532
72	€ <u>N</u> -5	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	575
73	N_}-{-}	6-ethoxy	Н	Н	-CH₂-	2-NH <sub>2</sub>	541

74	H <sub>3</sub> C =N - }	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	528
75	N_)-5	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	515
76	( ). S-	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	591
77	<u>ζ</u> , ζ	5-CI	Н	Н	-CH <sub>2</sub> -	2-NH₂	530
78	N	5-Cl	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	530
79	CI CI	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	548
80	N_}-}	5-CF <sub>3</sub>	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	565
81	N_N 5	Н	Н	Н	-CH₂-	2-NH <sub>2</sub>	497
82	CI CI	6,7-di-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	567
83	<b>₹</b> N-2	6,7-di-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	532
84	N=ZOH	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	530
85	CI Z	5-CF <sub>3</sub> ,7-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	617
86	N	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	529
87	O CH₃ N N	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	500
88	\$	Н	Н	H	-CH <sub>2</sub> -	2-NH <sub>2</sub>	485
89	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	Н	Н	Н	-CH₂-	2-NH <sub>2</sub>	489
90	S ~~	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	514
91	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	6-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	503

			· · · ·	T	011		ſ
92	<b>~</b>	5-F	Н	H	-CH <sub>2</sub> -	2-NH₂	503
93	Şs ₩	Н	Н	Н	-CH₂-	2-NH <sub>2</sub>	501
94	H <sub>3</sub> C O N	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	518
95	H₃C O N	5-CI	Н	Н.	-CH₂-	2-NH₂	534
96	\$\sigma_s\cdots	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	519
97	ON CH <sub>3</sub>	6,7-di-F	Н	Н	-CH₂-	2-NH₂	536
98	H <sub>3</sub> C ON N	5-Br	Н	Н	-CH₂-	2-NH₂	579
99	H <sub>3</sub> C ON N	6-ethoxy	Н	Н	-CH₂-	2-NH₂	544
100		5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	503
101	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	5-Br	Н	Н	-CH <sub>2</sub> -	2-NH₂	563
102	[	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	502
103	H <sub>3</sub> C ON N	5-CF <sub>3</sub>	Н	Н	-CH₂-	2-NH₂	568
104	H <sub>3</sub> C O N	5-CF <sub>3</sub> ,7-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	586
105	FOR S	5-F	Н	Н	-CH₂-	2-NH₂	598

	0-				<u> </u>		<del></del>
106	H₃C-	5-F	Н	Н	-CH₂-	2-NH₂	517
	C(CH <sub>3</sub> ) <sub>3</sub>		<del></del>	<del>                                     </del>	<u> </u>	0.111	
107	H <sub>3</sub> C-\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	5-F	Н	H	-CH₂-	2-NH₂	573
108	H <sub>3</sub> C ~~	5-F	Н	Н	-CH₂-	2-NH₂	517
109	CH <sub>3</sub> -S-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	483
110	CH <sub>3</sub> -CH <sub>2</sub> -S-	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	497
111	CH <sub>3</sub> -SO <sub>2</sub> -	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	515
112	<b>(</b> )—s−{}	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	545
113	H₃C H₃C S−Ş	5-F	H	Н	-CH <sub>2</sub> -	2-NH₂	511
114	F <sub>3</sub> C\_S-\{	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	551
115	$H_3C$ $H_3C$ $N$ $S$ $ \xi$	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	540
116	HS-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	469
117	CH <sub>3</sub> -S-	5-F	Н	2- CH₃	-CH <sub>2</sub> -	2-NH₂	497
118	CH <sub>3</sub> -S-	5-F	F	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	501
119	<b>₹</b> -N-ξ	5-F	Н	H	-CH₂-	2-NH₂	529
120	o_n−{	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	522
121	H <sub>3</sub> CO <sub>2</sub> S-N N-5	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	599
123	(T)-H-{}	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	528
124	F-\(\bar{\}\)-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	5-F	Н	Н	-CH₂-	2-NH₂	564
125	F CH <sub>3</sub>	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	578
126	SO <sub>2</sub> CH <sub>3</sub> F-√N-{ N-{	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	624
127	F-\( \bar{\bar{\bar{\bar{\bar{\bar{\bar{	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	546

400		<i>5</i>	<u> </u>	Н	CH	T 0 N 11 1	
128	F <sub>3</sub> CO <sub>2</sub> S-NN-}	5-F	H	П	-CH₂-	2-NH <sub>2</sub>	653
129	CH <sub>3</sub> -O-(CH <sub>2</sub> ) <sub>2</sub> - NH-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	510
130	H <sub>2</sub> N N-5	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	563
131	H <sub>3</sub> C N— {	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	480
132	CH <sub>3</sub> -O-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	467
133	CH <sub>3</sub> -CH <sub>2</sub> -O-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	481
134	CH <sub>3</sub> -O-(CH <sub>2</sub> ) <sub>2</sub> -O-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	511
135	(CH <sub>3</sub> ) <sub>2</sub> -CH-O-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	495
136	<b>─</b> -o-{	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	529
137	H <sub>2</sub> N-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	Н	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	511
138	<b>€</b> >-}	5-CF <sub>3</sub> ,7-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	582
139	€ <u>&gt;</u> -8	5-F	Н	Н	CH₃ −CH−	2-NH <sub>2</sub>	528
140	<u>ν</u> ξ	5-F	F	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	532
141	<u>ν</u> ξ	5-F	ОН	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	530
142	N_}-}	5-F	Н	Н	CH₃ -CH-	2-NH <sub>2</sub>	529
143	N_}-Ş	5-F	Н	Н	CH₃ -CH—	2-NH₂	529
144	<u>_</u> ~}	5-F	-CH₃	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	528
145	<u>ν</u> ξ	6-F	Н	Н	CH₃ -CH	2-NH <sub>2</sub>	528
146	Н	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	437
147	H <sub>3</sub> C-CH <sub>3</sub>	5-F	Н	H	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
148	H <sub>3</sub> C CH <sub>3</sub>	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	531
149	F <sub>3</sub> C CH <sub>3</sub>	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	585
L	~~	<u> </u>	L	L	L	ł	L

		r==	r:.	1	<u> </u>		
150	H <sub>3</sub> C CH <sub>3</sub>	5-F	Н	H	-CH₂-	2-NH <sub>2</sub>	549
151	CF <sub>3</sub>	5-F	Н	Н	-CH₂-	2-NH <sub>2</sub>	571
152	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	Н	F	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	514
153	(CH <sub>3</sub> ) <sub>2</sub> N-(CH <sub>2</sub> ) <sub>2</sub> - NH-	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	523
154	CH₃-S-	5-F	Н	н	CH₃ -CH-	2-NH <sub>2</sub>	497
155	<b>€</b> N-5	5-F	Н	2- CH <sub>3</sub>	-CH <sub>2</sub> -	2-NH <sub>2</sub>	528
156	<b>⟨</b> }_}}	5-F	Н	Н	-CH₂-	2-NH₂	514
157	<b>~</b> >~	5-F	Н	Н	-CH <sub>2</sub> -	3-NH₂	514
158	sN-}	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	589
159	\_N-\?	5-F	Н	Н	-CH <sub>2</sub> -	2-NH₂	520
160	CH₃CH₂O-	5-F	F	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	499
161	H <sub>3</sub> C CH <sub>3</sub> N-(CH <sub>2</sub> ) <sub>2</sub> -N-5 H <sub>3</sub> C	5-F	Н	Н	-CH₂-	2-NH₂	537
162	H3C-N_N-\$	5-F	Н	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	535
163	€ <u>N</u> -5	5-F	Н	5-OH	-CH₂-	2-NH <sub>2</sub>	530
164	€ <u>N</u> -5	5-F	F	Н	-CH <sub>2</sub> -	3-NH <sub>2</sub>	532
165	O_N-}	5-F	F	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	540
166	N_}-{>	5-F	Н	Н	-CH <sub>2</sub> -	3-NH₂	515

No.	R	R <sup>3</sup>	Z	R <sup>6</sup>	Physical Data MS (MH*)
167	<b>◯</b> -ş	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	502
168	-CH₂OCH₃	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	464
169	<b>←</b>	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	504
170	>~	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	460
171	(CH <sub>3</sub> ) <sub>2</sub> -CH-	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	462
172	H <sub>3</sub> C N	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	477
173	F—€}	Н	-CH₂-	2-NH <sub>2</sub>	514
174	F—Z	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	532
175	cı—{}-{-}	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	530
176	F-{}-{}-{}-{}-{}-{}-{}-{}-{}-{}-{}-{}-{}-	Н	-CH₂-	2-NH <sub>2</sub>	532
177	6- <del>-</del>	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	540
178	CI ~}	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	564
179	Стуру — Қ ОСН₃	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	526
180	H₃CH₂CO F	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	558
181	N=>	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	497
182	√Z- NH <sub>2</sub>	Н	-CH₂-	2-NH <sub>2</sub>	512
183	CI - S	H	-CH₂-	2-NH <sub>2</sub>	531

184	N}-&	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	498
185	<b>₹</b>	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	497
186	H <sub>3</sub> C =N -2	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	511
187	H <sub>3</sub> C O N	Н	-CH₂-	3-NH <sub>2</sub>	501
188	$\stackrel{\frown}{\sim}$	Н	-CH₂-	2-NH <sub>2</sub>	486
189		Н	-CH₂-	2-NH <sub>2</sub>	486
190	H <sub>3</sub> C O N N	Н	-CH₂-	2-NH <sub>2</sub>	501
191	(T)-3,	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	536
192	₽ <sub>N</sub>	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	547
193	Q Sz. N	Н	-CH <sub>2</sub> -	2-NH <sub>2</sub>	547
194	(CH₃)₃C , N , N	Н	-CH₂-	2-NH <sub>2</sub>	543
195	F-()-(0.N	Н	-CH₂-	2-NH <sub>2</sub>	581
196	H <sub>3</sub> C O.	F	-CH <sub>2</sub> -	2-NH <sub>2</sub>	519
197	H <sub>3</sub> C O.	Н	CH₃ -CH-	2-NH <sub>2</sub>	515
198	H <sub>3</sub> C O N	OH	-CH <sub>2</sub> -	2-NH <sub>2</sub>	517
199	H <sub>3</sub> C O N		-CH <sub>2</sub> -	2-NH <sub>2</sub>	577
200	<b>₹</b>	F	-CH <sub>2</sub> -	2-NH <sub>2</sub>	515

201	\$	F	-CH₂-	2-NH <sub>2</sub>	504
202	<u></u> _5N-5	Н	-CH₂-	3-NH <sub>2</sub>	497
203	F—₹	Н	-CH <sub>2</sub> -	3-NH <sub>2</sub>	532
204	<b>₹</b> N-2	F	-CH <sub>2</sub> -	3-NH <sub>2</sub>	515
205	F—Q−Şi F	F	-CH <sub>2</sub> -	3-NH <sub>2</sub>	550

No.	R	Physical Data MS (MH*)
206	-CH <sub>3</sub>	434
207	<b>₹</b>	497
208	F-{}-}	514
209	cı—⟨	530

$$\begin{array}{c|c}
R & N & N & R^2 \\
N & A & N & R^2
\end{array}$$

No.	R	R <sup>25</sup>	A	R <sup>3</sup>	R <sup>2</sup>	Physical Data MS (MH <sup>*</sup> )
210	N_>-5	5-CI	С	Н	$\xi \stackrel{N}{\longrightarrow} NH_2$	532
211	<u>₹</u> N	5-F	С	Н	$\xi \leftarrow NH_2$	515
212	N_√- ξ	5-CI	С	Н	ξ-(=N NH <sub>2</sub>	532

213	N_>-5	5-F	С	Н	ξ-(=N N NH <sub>2</sub>	516
214	<u></u> ~	Н	N	Н	ξ-(=N NH <sub>2</sub>	503
215	<b>○</b> -}	Н	N	Н	NH <sub>2</sub>	503
216	(CH₃) <sub>2</sub> CH-	Н	N	н	NH <sub>2</sub>	463
217	F——	5-F	С	Н	NH <sub>2</sub>	550
218	<b>€</b> N	5-F	С	Н	NH <sub>2</sub>	515
219	(=N)-3:	5-CI	С	Н	NH <sub>2</sub>	532
220	F—{}-};	6-CI	С	Н	NH2 NNH2	548
221	ν)—ζ·	5-F	С	Н	NH2	516
222	CI ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~	6-CI	С	Н	NH2	600
223	N}-Ş	5-CI	С	Н	NH <sub>2</sub>	532
224	€ <mark>N</mark> –§:	6-F	С	Н	ξ-(=N N NH <sub>2</sub>	515
225	ν <u>_</u> ν_ξ	Н	N	Н	ξ-(=N NH <sub>2</sub>	499
226	H <sub>3</sub> C O N	Н	N	Н	ξ—NH <sub>2</sub>	502
227		Н	N	Н	NH <sub>2</sub>	487

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228	CI <sub>N</sub> J <sub>5</sub> 5	Н	N	H	NH <sub>2</sub>	548
229	C N	Н	N	Н	ξ-(=N-NH <sub>2</sub>	548
230	~~} N}-}	Н	N	Н	NH <sub>2</sub>	499
231	H <sub>3</sub> C O N	Н	N	Н	NH <sub>2</sub>	502
232	(1) h	Н	N	Н	NH <sub>2</sub>	537
233	C N	Н	N	Н	NH <sub>2</sub>	548
234		Н	N	Н	NH <sub>2</sub>	541
235	CH₃CH₂O F	H	N	Н	ξ-(-N-NH <sub>2</sub>	559
236	<b>₹</b>	Н	N	H	ξ-(=N NH <sub>2</sub>	498
237	<b>€</b> }-};	5-F	С	F	ξ-(=N NH <sub>2</sub>	533
238	F-{\}-\{\}-\{\}-\{\}-\{\}-\{\}-\{\}-\{\}	5-F	С	Н	ξ-(=N-NH <sub>2</sub>	550
239	F—Ş−Ş	5-F	С	Н	ξ-(=N NH <sub>2</sub>	550
240	<u>~</u> _}-}	5-F	Ċ	Н	ξ-(=N N NH <sub>2</sub>	515
241	⟨ <u>N</u> _}-ξ	5-F	С	Н	ξ-(=N N NH <sub>2</sub>	516
242	<b>₹</b>	Н	С	Н	ξ-(=N NH <sub>2</sub>	497
243	(CH <sub>3</sub> ) <sub>2</sub> N-CH <sub>2</sub> -	Н	N	Н	ξ-(=N-NH <sub>2</sub>	478

244 H <sub>3</sub> C O		F	<del></del>		· · · · · · · · · ·		· · · · · · · · · · · · · · · · · · ·
246	244	H3C 0	5-F	С	Н	ξ-(N)NH2	519
247	245	Z N	Н	С	Н	ξ-(-N-NH <sub>2</sub>	501
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	246	H₃C O N	5,6-di-F	С	Н	ξ-(=N N NH <sub>2</sub>	537
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	247	<b>₹</b>	5-F	С	Н	ξ-\_N=_N	500
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	248	<u>√</u> -5	5,6-di-F	С	Н	ξ-(=N NH <sub>2</sub>	534
251 $N = N - \frac{1}{2}$ 5-F C F $\frac{1}{2} = N - \frac{1}{2}$ 534 252 $\frac{1}{N} - \frac{1}{2}$ 5-F C F $\frac{1}{2} = N - \frac{1}{2}$ 533 253 $\frac{1}{N} - \frac{1}{2}$ 5-F C F $\frac{1}{2} = N - \frac{1}{2}$ 568 254 $\frac{1}{N} - \frac{1}{2}$ 5-F C F $\frac{1}{2} = N - \frac{1}{2}$ 568 255 $\frac{1}{N} - \frac{1}{2}$ H N H $\frac{1}{2} = N - \frac{1}{2}$ 515 256 $\frac{1}{N} - \frac{1}{2}$ H C F $\frac{1}{2} = N - \frac{1}{2}$ 515 257 $\frac{1}{N} - \frac{1}{2}$ H N F $\frac{1}{2} = N - \frac{1}{2}$ 516 259 $\frac{1}{N} - \frac{1}{2}$ H N F $\frac{1}{2} = N - \frac{1}{2}$ 505	249	FO.	5-F	С	F	ξ-{ }-NH <sub>2</sub>	537
251 N $\rightarrow \xi$	250	<u>√</u> -5	5-F	С	F	ξ-(=N NH <sub>2</sub>	534
253 $F \longrightarrow -\frac{1}{2}$ 5-F C F $\frac{1}{2} \longrightarrow -\frac{1}{2}$ 568  254 $F \longrightarrow -\frac{1}{2}$ 5-F C F $\frac{1}{2} \longrightarrow -\frac{1}{2}$ 568  255 $\frac{1}{2} \longrightarrow -\frac{1}{2}$ H N H $\frac{1}{2} \longrightarrow -\frac{1}{2}$ 515  256 $\frac{1}{2} \longrightarrow -\frac{1}{2}$ H C F $\frac{1}{2} \longrightarrow -\frac{1}{2}$ 515  257 $\frac{1}{2} \longrightarrow -\frac{1}{2}$ H N F $\frac{1}{2} \longrightarrow -\frac{1}{2}$ 516  259 $\frac{1}{2} \longrightarrow -\frac{1}{2}$ H N F $\frac{1}{2} \longrightarrow -\frac{1}{2}$ 516	251		5-F	С	F	ξ-{ }-NH2	534
253	252	<u>~</u> }-{	5-F	С	F	ξ-(=N NH <sub>2</sub>	533
255	253	F—Ç}-Ş	5-F	С	F	ξ-{ }-NH <sub>2</sub>	568
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	254	F-{}-{}	5-F	С	F	ξ-( )-NH <sub>2</sub>	568
257 $H_3C$ $O$	255	C)	Н	N	H		487
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	256	€ <u>N</u> -8	Н	С	F	ξ-(=N NH <sub>2</sub>	515
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	257	H₃C O N	Н	С	F	ξ-(=N N NH <sub>2</sub>	519
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	258	N=){-}	Н	N	F	ξ-(=N NH <sub>2</sub>	516
260 $= N - \xi$ H N F $= k - N + N + N + N + N + N + N + N + N + N$	259	€_}	Н	N	Н	ξ-(=N NH <sub>2</sub>	505
	260	₹ <u></u>	Н	N	F	ξ-(-N-NH <sub>2</sub>	516

261	H³C O N	Н	N	F	ξ-(=N/NH <sub>2</sub>	520
262	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	5-F	С	H	ξ-(=N NH <sub>2</sub>	504
263	<b>€</b>	5-F	С	Н	ξ-(=N NH <sub>2</sub>	522
264		5-F	С	Н	ξ-(=N)-NH <sub>2</sub>	504
265	D}3	Н	N	H	$\xi = N$ $NH_2$	537
266	(CH <sub>3</sub> ) <sub>2</sub> N-CH <sub>2</sub> -	Н	N	F	ξ-(=N NH <sub>2</sub>	496
267	$\sqrt{2}$	Н	N	F	ξ-(=N)-NH <sub>2</sub>	505
268	CH₃CH₂-O-	5-F	С	Н	ξ-(=N NH <sub>2</sub>	482
269	CH <sub>3</sub> -S-	5-F	С	Н	ξ-(=N NH <sub>2</sub>	484
270	CH₃CH₂-O-	5-F	С	F	$\xi = N \longrightarrow NH_2$	500
271	<b>्र</b> ेर	Н	N	F	$\xi = N \longrightarrow NH_2$	555
272	C N	H	N	F	ξ-(=N)-NH <sub>2</sub>	566
273	<u>N</u> =>-}-}	Н	N	Н	ξ-{=N NH <sub>2</sub>	498
274	<u></u>	5,6- di-F	С	F	ξ-(=N NH <sub>2</sub>	551
275	<b>√</b> N−}	5-F	С	F	ξ-{=N NH <sub>2</sub>	541
276	Q_N-{	5-F	С	Н	ξ-(=N NH <sub>2</sub>	523
277	<b>₹</b>	5-F	С	Н	ξ-\(\big _N\) H <sub>2</sub> N	514
<u> </u>	L	Ь		Ц	1	

278	<b>₹</b>	5-F	С	Н	ξ-√n N∞N	539
279	H <sub>3</sub> C — CH <sub>3</sub>	Н	N	Н	ξ-(-N-NH <sub>2</sub>	515
280	H₃C √O	Н	N	Н	ξ-(-N-NH <sub>2</sub>	501
281	\$	Н	N	F	ξ-(=N)-NH <sub>2</sub>	505
282	J. S	Н	N	Н	ξ-(=N-NH <sub>2</sub>	536
283	€)-}	Н	N	F	ξ-(=N-NH <sub>2</sub>	523
284	<b>₹</b>	5-F	С	F	ξ-\(\bigc\) H <sub>2</sub> N	532
285	H <sub>3</sub> C O	Н	N	Н	ξ-(-N-NH <sub>2</sub>	501
286	F-{\(\)\{\}	Н	N	Н	ξ-(=N NH <sub>2</sub>	533
287	N=N N_}-&	Н	N	F	$\xi - N - NH_2$	517
288	CZ, N	Н	N	Н	ξ-(=N NH <sub>2</sub>	548
289	F—————————————————————————————————————	Н	N	Н	ξ-(=N N-NH <sub>2</sub>	533
290	CH <sub>3</sub> S-	5-F	С	F	ξ-(=N NH <sub>2</sub>	502
291	<b>€</b> N-}-}	Н	. N	F	ξ-{\bigcirc_N}-NH2	515
292	<b>€</b> N-5	5-F	С	F	ξ-(	532
293	<u>₹</u> N{	5-F	С	Н	ξ-(N-NH <sub>2</sub>	514
294	<b>₹</b>	Н	N	Н	$\xi$ $NH_2$	497

295	(CH <sub>3</sub> ) <sub>2</sub> N-	5-F	С	F	) /=N	499
250	(0.13/2.1			<u> </u>	$\xi = N$ $NH_2$	400
296	CH₃CH₂-S-	5-F	С	F	ξ-(=N N-NH <sub>2</sub>	516
297	CH₃-O-	5-F	С	F	ξ-(=N-NH <sub>2</sub>	486
298	<b>₹</b>	Н	N	Н	ξ-{\(\big \) OCH3	512
299	<b>€</b> N-5	Н	N	F	ξ- <b>(</b> N) OCH <sub>3</sub>	530
300	<b>₹</b>	5-F	С	F	ξ- <b>(</b> N-och₃	547
301	<b>₹</b>	5-F	С	Н	ξ-√N-осн₃	529
302	<u></u>	5-F	С	Н	ξ-{\(\)_F	517
303	<u>~</u> N-}-}	5-F	С	F	}-{\big _{F}^{N}}	535
304	<u>₹</u> N-}	Н	N	Н	ξ-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	551 _
305	F—Q	Н	N	F	ξ-(-N-NH <sub>2</sub>	551
306	₹ <u></u>	5-F	С	Н	ξ-{¬»	500
307	₹ <u>N</u> -Ş	5-F	С	Н	ξ-(N	500
308	<u>~</u> N};	5-F	С	F	ξ-N-NH <sub>2</sub> H <sub>3</sub> C	547
309	(CH <sub>3</sub> CH <sub>2</sub> ) <sub>2</sub> N-	5-F	С	F	ξ-(=N-NH <sub>2</sub>	527
310	₹ <u></u>	Н	N	Н	ξ-(NH ο	498
311	<u>₹</u> N-ξ	Н	N	F	ξ-(NH ο	516
312	€ <u>N</u> -5	5-F	С	Н	ξ-(_NH o	515

	<u> </u>					
313	<b>₹</b>	5-F	С	F	ξ-(_NH o	533
314	H <sub>3</sub> C N-5	5-F	С	F	ξ-(=N-NH <sub>2</sub>	569
315	CH <sub>3</sub> -S-	Н	N	F	ξ-(=N N-NH <sub>2</sub>	485
316	CH₃CH₂-O-	Н	N	F	ξ-(=N N NH <sub>2</sub>	483
317		Н	N	F	ξ-(=N NH <sub>2</sub>	566
318	<b>₹</b>	Н	N	F	<b>₹~</b>	489
319	~~ >	Н	N	F	₹	489
320	_ <b>N</b>	Н	N	F	ξ-{ <sup>S</sup> }	505
321	~~~	Н	N	F	δ-√S	505
322	<b>€</b> }-}	5-F	С	F	ξ-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	533
323	€ <u>n</u> -5	Н	N	F	ξ-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	516
325	F-\$	Н	N	F	δ-√S	540
325	0_N−}	Н	N	F	ξ-(=N NH <sub>2</sub>	524
326	(CH <sub>3</sub> ) <sub>2</sub> CH-O-	5-F	С	F	ξ-(=N NH <sub>2</sub>	514
327	<b>₹</b>	Н	N	F	}	506
328	<b>₹</b>	Н	N	F	₹ <u>~</u> "	488
329	<b>₹</b>	Н	N	F	δ, N-N	489
330	<u>₹</u> N§	Н	N	F	S-√N=N N=N	507
331	<u></u>	Н	N	F	ξ-{S}-scH₃	551

		т	T	1	A)	<u> </u>
332	<u></u>	Н	N	F 	δ-\\_s	506
333	<b>₹</b> }-}	Н	N	F	H <sub>3</sub> C N O H <sub>3</sub> C	518
334	€ <u>n</u> -}-}	Н	N	F	ζ,—(CH <sub>3</sub>	504
335	CH₃-O-	Н	N	F	ξ-(-N-NH <sub>2</sub>	464
336	<b>₹</b>	Н	N	F	ς√ <sub>N</sub> −0	491
337	<b>₹</b> }-}	Н	N	F	₹ NC(O)CH3	563
338	<u>₹</u> "–}	5-F	С	Н	H <sub>3</sub> C CH <sub>3</sub>	545
339	<u>~</u> N}	5-F	С	F	ξ-N-NH <sub>2</sub>	533
340	<u>₹</u> N-5	Н	N	F	ξ-(O)-NH <sub>2</sub>	518
341	<u>~</u> N>	5-F	С	Н	ξ_O_NH <sub>2</sub>	535
342	<b>€</b> }–}	Н	N	F	Z√NS CH₃	520
343	F—{	6-CI	С	Н	ξ-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	548
345	<u></u>	Н	N	Н	ξ-(N-NH <sub>2</sub> )	503
346	(CH <sub>3</sub> ) <sub>2</sub> -CH-	Н	N	Н	ξ-(NNNH <sub>2</sub> NH <sub>2</sub>	463

No.	R <sup>3</sup>	R <sup>2</sup>	Physical Data MS (MH*)
347	Н	ξ-(=N NH <sub>2</sub>	489
348	F	ξ-(N <sub>NH2</sub>	506
349	F	ξ-(_N NH <sub>2</sub>	488
350	F	ξ-(=N NH <sub>2</sub>	507
351	F	ξ-(N NH <sub>2</sub>	506

$$R^{1}$$

No.	R <sup>1</sup> -X-	Z	R <sup>3</sup>	R <sup>2</sup>	Physical Data MS (MH <sup>+</sup> )
352		-CH <sub>2</sub> -	Н	ξ-(N <sub>NH₂</sub>	509
353		-CH₂-	H	ξ-(N <sub>NH₂</sub>	510
354		-CH₂-	Н	ξ-√N NH <sub>2</sub>	523

355	F N S	-CH₂-	H .	ξ-(N <sub>NH<sub>2</sub></sub>	532
356		-CH₂-	Н	ξ-(N NH <sub>2</sub>	496
357	CH <sub>2</sub> ) <sub>2</sub> OCH <sub>2</sub> CH <sub>3</sub>	-CH₂-	Н	Ş-(N NH₂	506
358	F H J H J Y	-CH₂-	Н	ξ-(N <sub>NH<sub>2</sub></sub>	542
359	H SY	-CH <sub>2</sub> -	Н	ξ-(N NH <sub>2</sub>	451
360		-CH₂-	Н	ξ-(\_\N NH <sub>2</sub>	537
361		-CH₂-	Н	ξ-(\_N NH <sub>2</sub>	495
362	NN N CH₂CF₃	-CH <sub>2</sub> -	Н	Ş-(N NH <sub>2</sub>	501
363		-CH <sub>2</sub> -	Н	ξ-(N NH <sub>2</sub>	510
364	F N N N N N N N N N N N N N N N N N N N	-CH₂-	Н	ξ-(-N-NH <sub>2</sub>	533
365	N N N N N N N N N N N N N N N N N N N	-CH <sub>2</sub> -	Н	ξ-(_N NH <sub>2</sub>	420

366	<b>₹</b>	-CH <sub>2</sub> -	Н	₹-{\n'\	449
	H₃CO N '		1,0	NH <sub>2</sub>	
367	J. J.	-CH <sub>2</sub> -	Н	ξ-N-NH <sub>2</sub>	497
	N S			_N	
368	F N S	-CH₂-	Н	ξ-(N	533
	τι CH₂CF₃			NH <sub>2</sub>	
369	CI N N	-CH₂-	Н	ξ-(N <sub>NH<sub>2</sub></sub>	487
370	CINY S	-CH <sub>2</sub> -	Н	ξ-( N	509
				NH <sub>2</sub>	
371	(I)	-CH <sub>2</sub> -	Н	\$-{_n	433
	N N			NH <sub>2</sub>	
372	F <sub>3</sub> C N = \{	-CH <sub>2</sub> -	Н	ξ-(_n	504
	~ 5 .			NH <sub>2</sub>	
373	(CIS→ §	-CH <sub>2</sub> -	H	ξ-(N NH <sub>2</sub>	436
374	CI N	-CH <sub>2</sub> -	н	ξ-√\n	472
	<b>⋄</b> -0 ,			NH <sub>2</sub>	
375	(Cts)→{	-(CH <sub>2</sub> ) <sub>3</sub> -	Н	ξ-(N NH <sub>2</sub>	464
376		-CH <sub>2</sub> -	Н	ξ-√"N	544
	in i			NH <sub>2</sub>	
	cı				
377	Ĉ <sub>N</sub>	-CH <sub>2</sub> -	F	\$-(_n	562
	η η η η η η η η η η η η η η η η η η η			NH <sub>2</sub>	
	CI	·			
<u> </u>			L	<del></del>	L

No.	R	M <sup>1</sup>	Y	R <sup>2</sup>	Physical Data MS (MH*)
378	<u>_</u> _5	СН	-CH <sub>2</sub> -	Ş-{\_N NH <sub>2</sub>	500
379	N}-5	N	-NH-	ξ-(=N NH <sub>2</sub>	502
380	<b>\( \)</b>	N	-NH-	ξ-(NH <sub>2</sub>	490
381	$\searrow$	N	-NH-	ξ-(=N NH <sub>2</sub>	494
382	<u></u>	N	-NH-	ξ-(=N NH <sub>2</sub>	501
383	₹ <mark>N</mark> -Ş	N	-NH-	ξ-\_N NH <sub>2</sub>	500

384:

MS: 528 (MH<sup>+</sup>)

385:

386:

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MS 583 (MH<sup>+</sup>)

5 Step 1:

A solution of P7-1 (2.3 g, 8.9 mmol) in CH<sub>2</sub>Cl<sub>2</sub>-DMF (1:1, 50 ml) was treated with picolinic acid N-oxide (1.5 g, 10.6 mmol), EDCI (2.6 g, 13.3 mmol) and HOBT (1.8 g, 13.3 mmol). The mixture was stirred at 70 °C overnight. The reaction mixture was concentrated, diluted with EtOAc, washed three times with 5% aqueous NaOH, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated. Flash chromatography (50% EtOAc/hexane) provided 388A (2.5 g, 74%).

## Step 2:

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In a manner similar to that described in Preparation 5, Step 4, compound **388A** was converted to compound **388B**.

## Step 3:

A solution of 388B (0.66 g, 2.2 mmol) in DMF (15 ml) was treated with 5C (0.62 g, 2.5 mmol), 1-propanephosphonic acid cyclic anhydride (3.3 ml, 11.2 mmol, 50 wt. % in EtOAc) and N-ethylmorpholine (1.4 ml, 10.7 mmol). The mixture was stirred at 50 °C for 3h. The reaction mixture was concentrated and diluted with EtOAc. The solution was washed three times with 5% aqueous NaOH, dried over Na<sub>2</sub>SO<sub>4</sub>, concentrated and subjected to flash chromatography (10% 2N NH<sub>3</sub>-CH<sub>3</sub>OH/EtOAc). The material was then taken up in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) and treated with 4 M HCl-dioxane (4 ml). After stirring overnight at 20 °C, the reaction was carefully basified with 10% aqueous NaOH and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub>, concentrated and subjected to flash chromatography (30% 2N NH<sub>3</sub>-CH<sub>3</sub>OH/EtOAc) to provide **388C** as a white solid (0.08g, 10%). Step 4:

In a manner similar to that described in Example 5, Step 5, compound 388C was converted to Example 388.

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### Example 389

## Step 1:

To a stirred, cloudy solution of 389A (300 mg, 1.14 mmol) in THF (15 ml) were added a solution of 389B (292 mg, 1.37 mmol) in THF (1 ml), followed by NaBH(OAc)<sub>3</sub> (483 mg, 2.28 mmol). After stirring at RT for 39 h, TLC revealed the presence of unchanged starting materials in the cloudy white reaction suspension. Therefore, another quantity of NaBH(OAc)<sub>3</sub> (242 mg, 1.14 mmol) was added and stirring at RT continued for a total of 113 h. The reaction mixture was then filtered and collected solids washed thoroughly with CH<sub>2</sub>Cl<sub>2</sub>. The combined filtrate and washings were stripped of solvent under vacuum, and the residue was partitioned between EtOAc (60 ml) and a solution consisting of water (2.5 ml), 2M Na<sub>2</sub>CO<sub>3</sub> (6.5 ml) and 6N NaOH (5

ml). The aqueous layer was further extracted with EtOAc (3 x 15 ml). The combined extracts were washed with brine (5 ml) and dried over anhydrous MgSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum. The residue was purified by silica gel flash chromatography (EtOAc/hexanes = 1:1) to obtain **389C** as a mixture of colorless gum and white foam (368 mg, 70%), homogeneous to TLC, which solidified upon standing. ES-MS: 461 (MH $^+$ ; 100%).

Step 2:

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To a stirred, ice-cold solution of **389C** (358 mg, 0.777 ml) in  $CH_2Cl_2$  (7 ml) was added via syringe cold, neat TFA (576 microliters, 886 mg, 7.77 mmol). The resultant solution was stirred in an ice-water bath for 30 min, then at RT for 29.5 h. Volatiles were removed under vacuum, and the gummy residue was triturated (magnetic stirrer) with  $Et_2O$  (35 ml) for 16 h. Filtration and washing with  $Et_2O$  yielded the bistrifluoroacetate salt of **389D** as a white powder (449 mg, 98%).

<u>Step 3</u>:

To a stirred suspension of 389D (100 mg, 0.170 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 ml) was added Et<sub>3</sub>N (47.4 microliters, 34.4 mg, 0.340 mmol), whereupon all solids dissolved. To the stirred solution were then added 5G (25.1 mg, 0.204 mmol), followed by NaBH(OAc)<sub>3</sub> (72.1 mg, 0.340 mmol). After stirring at RT for 66 h, TLC revealed the presence of unchanged starting materials in the light yellow reaction suspension. Therefore, another quantity of NaBH(OAc)<sub>3</sub> (72.1 mg, 0.340 mmol) was added and stirring at RT continued for a total of 90 h. The reaction mixture was then filtered and collected solids washed thoroughly with CH<sub>2</sub>Cl<sub>2</sub>. The combined filtrate and washings were stripped of solvent under vacuum, and the residue was partitioned between EtOAc (20 ml) and a solution consisting of water (0.6 ml), 2M Na<sub>2</sub>CO<sub>3</sub> (1.5 ml) and 6N NaOH (1.2 ml). The aqueous layer was further extracted with EtOAc (3 x 5 ml). The combined extracts were washed with brine (2 ml) and dried over anhydrous MqSO<sub>4</sub>. Drying agent was removed by filtration, and the filtrate was concentrated under vacuum. The residue was purified by preparative TLC (silica gel;  $CH_2CI_2/CH_3OH/conc.$   $NH_4OH = 90:9:1)$  to obtain the title compound as a light beige foam (36 mg, 45%). FABMS: 468 (MH<sup>+</sup>; 100%).

Using procedures similar to those described above in Examples 1-6 and 388-389, the following compounds were prepared:

Ex.	Structure	Mass Spec (M+H)
390	N N N N N N N N N N N N N N N N N N N	533 (ESMS)
391	N N N N N N N N N N N N N N N N N N N	518 (ESMS)
392	N N N N N N N N N N N N N N N N N N N	535 (ESMS)
393	N S CH <sub>3</sub>	520 (ESMS)
394	N-9=0	592 (FAB)
395	N O O CH <sub>3</sub> N O CH <sub>3</sub> O CH <sub>3</sub>	670 (FAB)

396	B H-C	528
	N N-CH	(ESMS)
	N N N N N N N N N N N N N N N N N N N	
	N N	
397		491
	N N=N	(ESMS)
	N N N	+
	N	
398		470
	N N N	(ESMS)
	N N N NH	
	N	
222		100
399		488
		(ESMS)
	N S	
	N	
400		487
	NY (NY)	(ESMS)
	N	
401		471
10.		(ESMS)
		(2011.0)
	N N	
402		487
	N N	(ESMS)
	N N S	
L		ــــــــــــــــــــــــــــــــــــــ

403	0	471
403		
		(ESMS)
	N .	
404		489
	>=N	(ESMS)
	N	
	N N S	
	N=N'S	
405		506
	N=N	(ESMS)
	$\dot{N}$	
	)     V	
406		505
	N=\N=\N	(ESMS)
	N N NH	
	F N N	
407		522
	N=\rightarrow\n	(ESMS)
	N N N S	
	F	
	Ü Ė	500
408	>	522
	N=\(\begin{array}{c} N \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \	(ESMS)
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	F N S	
400	Ü F	506
409	<u>_</u> >	
	N=(N)	(ESMS)
	F N N F	
	l ö <sup>F</sup>	L

410		523
	N= N	(ESMS)
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	F N N N	
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411	<b>_</b> _>	524
	N=\_N	(ESMS)
	N F S	
412	0 -	501
		(ESMS)
		:
	N	
413	0	490
		(ESMS)
	N N N N N N N N N N N N N N N N N N N	(==:)
414		473
414	N=N	(ESMS)
	\	(LONO)
	N	
415		488
		(ESMS)
	N N V	
	E	
416		487
	N N N	(ESMS)
	N NH	
	<u> </u>	<u> </u>

417	0	504
	N CN	(ESMS)
	N. Le»	
	N N	
418		504
410		
		(ESMS)
	N N V	
	F	
419		488
-		(ESMS)
	F.	
420		505
	N S	(ESMS)
	N N N	!
421	<u> </u>	506
	N=N N=N S	(ESMS)
422	F OF N NH.	526
	$N$ $NH_2$	(FAB)
	N N N N N N N N N N N N N N N N N N N	
L	<u> </u>	

423	I S	518
	N N N	(ESMS)
1		
424	9	585
		(FAB)
		(17.0)
	NH NH CH <sub>3</sub>	
425	F	591
	\\\\\\	(ESMS)
1	N=(	(LONG)
	N N N N N N N N N N N N N N N N N N N	
	N NH <sub>2</sub>	
	NH CH <sub>3</sub> N NH <sub>2</sub>	
	o <sup>s</sup> ~o	
426	9	499
	N NH2	(ESMS)
	N N N N N N N N N N N N N N N N N N N	
	N	
	, N	
427		516
	N NH2	(ESMS)
		(LONO)
	"	
428		546
720		·
	N F	(ESMS)
	N N	
	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	
	N-CH <sub>3</sub>	
	H₃Ć	L

429		498
,,_0		(ESMS)
	N N	(,
	$N \sim NH_2$	
430	0-N+	514
	N	(ESMS)
	F.N. A A A	
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431		674
	YN NN N	571
	N N N N N N N N N N N N N N N N N N N	(ESMS)
	) <u>—(</u>	
400	N N	F00
432	OF F	589
		(ESMS)
	N N N N N N N N N N N N N N N N N N N	
	⟨◯N	
433		573
	N NN ~	(ESMS)
	N N N N N N N N N N N N N N N N N N N	
	(N	<b>***</b>
434	S S S S S S S S S S S S S S S S S S S	591
		(ESMS)
	N N S N	
	N .	
435	CH <sub>3</sub> N NI NIL	512
	N NH <sub>2</sub>	(ESMS)

436	CH <sub>3</sub> U F	530
, , ,		(ESMS)
		(LONIO)
437		483
	N N N	(ESMS)
	N	
438		484
	N-CH <sub>3</sub>	(ESMS)
	N N N CI 3	
	N	
439	O F	502
		(ESMS)
	N N-CH <sub>3</sub>	
	N	
440	H₃Ç Ü F	499
	s N NH <sub>2</sub>	(FAB)
	) \ <u>1</u>	
441		471
	N N NH	(ESMS)
	N N N N N N N N N N N N N N N N N N N	
	Z <sup>N</sup>	
442		488
772		(ESMS)
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444	0 MS) 8 MS)
444	0 MS) 8 MS)
444	MS) 8 MS)
445 NH (ESM ) NH	MS) 8 MS)
445 NH (ESM ) NH	MS) 8 MS)
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610	H <sub>3</sub> C OF	501
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611	H <sub>3</sub> C O	547
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612	H₃C, Q	531
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613	H <sub>3</sub> C O	543
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614	H₃Ç Ç	558
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621		511
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622	<b>O</b> C	496
		(ESMS)
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623	0	510
		(ESMS)
		(20110)
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	H <sub>3</sub> C	
624	O <sub>F</sub>	503
•	H <sub>3</sub> C-s, N	(ESMS)
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	CI	
625	O F	518
	H <sub>3</sub> C-s N NH <sub>2</sub>	(ESMS)
	CI	
626		505
	N	(ESMS)
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627		498
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628		485
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629	O O	481
		(ESMS)
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630	H₃C O	499
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631	N P N	499
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632	H <sub>3</sub> C O N NH <sub>2</sub>	514
	S N Y N N N N N N N N N N N N N N N N N	(ESMS)
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633	CI H₃Ç ÜF	517
	S NN N	(ESMS)
		(LOMO)
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	ĆI	
634	H <sub>3</sub> C	532
	S N N NH2	(ESMS)
	N N N N	
635	CI	488
		(ESMS)
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	Ŭ <sup>N</sup>	
636		518
	NN	(ESMS)
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637	H₃C OF	451
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638	F	537
	s,	(MH+)
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	N O O	
		470
639		472
		(MH+)
	( ) N	
640	OF.	519
		(MH+)
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	F	
641	0	487
		(MH+)
	N N N NH	
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642		516
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	N N N NH	
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	H₃C−\$=O O	
643		503
		(MH+)
	N N N NH	
		:
	CI	
<b></b>	<u> </u>	

644		484
644		
	H <sub>3</sub> C-S N N NH <sub>2</sub>	(ESMS)
	$^{N}$ $\stackrel{N}{\sim}$ $^{N}$ $\stackrel{N}{\sim}$ $^{N}$	
	F	
645	O F	503
	H <sub>3</sub> C- <sub>S</sub> NNH <sub>2</sub>	
	1 1 1 1 1 U	(ESMS)
	F	400
646	O N N/I	498
	H <sub>3</sub> C S N N NH <sub>2</sub>	(ESMS)
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	)	
	F	
647	O F	516
	A N NHO	
		(ESMS)
640	F	400
648		468
	H <sub>3</sub> C <sub>-O</sub> N NH <sub>2</sub>	(ESMS)
	$N \sim N \sim N$	
1	N, M A A A	
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649	F	486
649		
649	H <sub>3</sub> C-O N NH <sub>2</sub>	486 (ESMS)
649	H <sub>3</sub> C-O N NH <sub>2</sub>	
649	H <sub>3</sub> C <sub>2</sub> ONNNH <sub>2</sub> NNH <sub>2</sub>	
	H <sub>3</sub> C-O N N NH <sub>2</sub>	(ESMS)
649	H <sub>3</sub> C <sub>2</sub> O <sub>N</sub> N <sub>N</sub> NH <sub>2</sub>	(ESMS)
	H <sub>3</sub> C- <sub>S</sub> N N N N N N N N N N N N N N N N N N N	(ESMS)
	H <sub>3</sub> C <sub>2</sub> O <sub>N</sub> N <sub>N</sub> NH <sub>2</sub>	(ESMS)
	H <sub>3</sub> C-S N	(ESMS)
650	H <sub>3</sub> C-O N N N NH <sub>2</sub> H <sub>3</sub> C-S N N N N N N N N N N N N N N N N N N N	(ESMS) 469 (ESMS)
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650	H <sub>3</sub> C <sub>-</sub> O <sub>N</sub> N N <sub>N</sub> N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N N <sub>N</sub> N 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652	0	483
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653	₽ F	501
	H <sub>3</sub> C S N N N	(ESMS)
9	N N N	
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654	0	453
	H <sub>3</sub> C-O N N	(ESMS)
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655	F	171
000	H <sub>3</sub> C <sub>2</sub> O N N N N	471
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	F	
656	O F	468
		(ESMS)
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657	0	450
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658	0	530
	H <sub>3</sub> C N NH	(ESMS)
	N N N	
CEO	H₃CO₂Ś	
659	CH <sub>3</sub>	
	N <sup>3</sup> -N <sup>3</sup>	
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660	ÇH₃	453
	$N_{\text{o}}$	(FAB)
		(-1)
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661	H <sub>3</sub> C-S N NH <sub>2</sub>	470
	N N N N N N N N N N N N N N N N N N N	(FAB)
662	H <sub>3</sub> C-Ş NNN	455
		(FAB)
663	F H₃C Q	497
	s N	(ESMS)
	N N N N N N N N N N N N N N N N N N N	,
664	H <sub>3</sub> C O	481
	o Chy Ly	(FAB)
	F	
664A	H₃Ç PF	499
	\$ NNN	(FAB)
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4-[[4-[2-(5-methyl-3-isoxazolyl)-3H-imidazo[4,5-b]pyridine-3-yl]-1-(4-piperidinylcarbonyl)piperidine (0.99 g, 2.51 mmoles) and pyridazine 4-carboxaldehyde (0.35 g, 3.26 mmoles) were stirred at RT in dry CH<sub>2</sub>Cl<sub>2</sub> (25 ml) containing activated 3Å molecular sieves (6.5 g). After 5 h, triacetoxy borohydride (3.2 g, 15 mmoles) was added and the mixture was stirred for 70 h. The mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> and the solid filtered through a pad of Celite. The filtrate was stirred for 20 min. with saturated aqueous NaHCO<sub>3</sub>, then separated, washed with brine, and dried over anahydrous Na<sub>2</sub>SO<sub>4</sub>. The reaction mixture was purified by preparative TLC. The plates were eluted with EtOAc:Hexanes:CH<sub>3</sub>OH(NH<sub>3</sub>) (75:20:5). Extraction of the bands with 13% CH<sub>3</sub>OH(NH<sub>3</sub>)/EtOAc gave a mixture of Example 665 and Example 496. Example 658: MS (M+H): 423.

In a similar manner, using 4-[[4-[2-(methylthio)-3H-imidazo[4,5-b]pyridine-3-yl]-1-(4-piperidinylcarbonyl)piperidine (0.88 gr.;2.44 mmoles), pyridazine 4-carboxaldehyde (0.34 g, 3.18 mmoles), and triacetoxy borohydride, a mixture of Example 666 and Example 495 was prepared:

# 20 General Procedure for H<sub>3</sub>-Receptor Binding Assay

The source of the  $\rm H_3$  receptors in this experiment was guinea pig brain. The animals weighed 400-600 g. The brain tissue was homogenized with a solution of 50 mM Tris, pH 7.5. The final concentration of tissue in the homogenization buffer was 10% w/v. The homogenates were centrifuged at 1,000 x g for 10 min. in order to remove clumps of tissue and debris. The resulting supernatants were then centrifuged at 50,000 x g for 20 min. in order to sediment the membranes, which were next

washed three times in homogenization buffer (50,000 x g for 20 min. each). The membranes were frozen and stored at -70°C until needed.

All compounds to be tested were dissolved in DMSO and then diluted into the binding buffer (50 mM Tris, pH 7.5) such that the final concentration was 2  $\mu$ g/ml with 0.1% DMSO. Membranes were then added (400  $\mu$ g of protein) to the reaction tubes. The reaction was started by the addition of 3 nM [ $^3$ H]R- $\alpha$ -methyl histamine (8.8 Ci/mmol) or 3 nM [ $^3$ H]N $^{\alpha}$ -methyl histamine (80 Ci/mmol) and continued under incubation at 30°C for 30 min. Bound ligand was separated from unbound ligand by filtration, and the amount of radioactive ligand bound to the membranes was quantitated by liquid scintillation spectrometry. All incubations were performed in duplicate and the standard error was always less than 10%. Compounds that inhibited more than 70% of the specific binding of radioactive ligand to the receptor were serially diluted to determine a Ki (nM).

## General Procedure for rHu H<sub>3</sub> Binding Assay

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[<sup>3</sup>H]N<sup>c</sup>-methylhistamine (82 Ci/mmole) was obtained from Dupont NEN. Thioperamide was obtained from the Chemical Research Department, Schering-Plough Research Institute.

HEK-293 human embryonic kidney cells stably expressing the human histamine H<sub>3</sub> receptor were cultured in Dulbecco's modified Eagle's medium/10% fetal calf serum/penicillin (100 U/ml)/streptomycin (100 μg/ml)/Geneticin (0.5 mg/ml) at 37° C in a humidified 5% CO<sub>2</sub> atmosphere. Cells were harvested between passages five and twenty at 37° C in 5 mM EDTA/Hank's balanced salt solution and processed for membrane preparation. After low-speed centrifugation, ten min at 1000 xg, they were put into ten volumes of ice-cold buffer and disrupted with a Polytron (PTA 35/2 tip, 30 sec at setting 6). After subsequent low-speed centrifugation, supernatant was centrifuged ten min at 50,000 xg. The high-speed pellet was resuspended in the original volume of buffer, a sample was taken for protein assay (bicinchoninic acid, Pierce) and the suspension was centrifuged again at 50,000 xg. Membranes were resuspended at 1 mg of protein/ml of buffer and frozen at -80° C until use.

Membrane (15 μg of protein) was incubated with 1.2 nM [<sup>3</sup>H]N<sup>α</sup>-methyl-histamine, without or with inhibitor compounds, in a total volume of 200 μl of buffer. Nonspecific binding was determined in the presence of 10<sup>-5</sup> M thioperamide. Assay mixtures were incubated for 30 min at 30° C in polypropylene, 96-well, deep-well

plates, then filtered through 0.3% polyethylenimine-soaked GF/B filters. These were washed three times with 1.2 ml of 4° C buffer, dried in a microwave oven, impregnated with Meltilex wax scintillant and counted at 40% efficiency in a Betaplate scintillation counter (Wallac).

 $IC_{50}$  values were interpolated from the data or were determined from curves fit to the data with Prism nonlinear least squares curve-fitting program (GraphPad Software, San Diego, CA).  $K_i$  values were determined from  $IC_{50}$  values according to the Cheng and Prusoff equation.

In these assays, compounds of formula I have a  $K_i$  within the range of about 0.1 to about 600 nM. Preferred compounds of formula I have a  $K_i$  within the range of about 0.1 to about 100 nM. More preferred compounds of formula I have a  $K_i$  within the range of about 0.1 to about 20 nM.

Representative compounds of the present invention tested according to the above procedures have the following Ki values:

		<del></del>
Ex.	Receptor	$K_{i}$
	Source	
1	rHu	1
3	Guinea pig	13
5	rHu	9
13	Guinea Pig	27
54	Guinea Pig	30
71	Guinea Pig	1
94	Guinea Pig	1
109	rHu	1
120	Guinea Pig	0.3
165	rHu	2
170	Guinea Pig	0.5
173	Guinea Pig	0.4
195	Guinea Pig	10
211	Guinea Pig	7
254	Guinea Pig	13
269	rHu	4
270	rHu	4
281	rHu	4
290	rHu	3
290	rHu	3
297	rHu	4
297	rHu	4
315	rHu	5
316	rHu	5
316	rHu	5 5 2
326	rHu	2

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Ex.	Receptor	Ki
	Source	
335	rHu	12
388	rHu	30
423	rHu	5
442	rHu	1
449	rHu	1
459	rHu	4
460	rHu	4
468	rHu	10
493	rHu	1
502	rHu	7
512	rHu	2
547	rHu	14
552	rHu	4
557	rHu	19
571	rHu	2
574	rHu	2
577	rHu	44
588	rHu	6
592	rHu	9
595	rHu	41
598	rHu	17
608	rHu	1
618	rHu	9
619	rHu	2
625	rHu	10
628	rHu	4

In this specification, the term "at least one compound of formula I" means that one to three different compounds of formula I may be used in a pharmaceutical composition or method of treatment. Preferably one compound of formula I is used. Similarly, "at least one H<sub>1</sub> receptor antagonist " means that one to three different H<sub>1</sub> antagonists may be used in a pharmaceutical composition or method of treatment. Preferably, one H<sub>1</sub> antagonist is used.

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For preparing pharmaceutical compositions from the compounds described by this invention, inert, pharmaceutically acceptable carriers can be either solid or liquid. Solid form preparations include powders, tablets, dispersible granules, capsules, cachets and suppositories. The powders and tablets may be comprised of from about 5 to about 95 percent active ingredient. Suitable solid carriers are known in the art, e.g. magnesium carbonate, magnesium stearate, talc, sugar or lactose. Tablets, powders, cachets and capsules can be used as solid dosage forms suitable for oral administration. Examples of pharmaceutically acceptable carriers and methods of manufacture for various compositions may be found in A. Gennaro (ed.), *The Science and Practice of Pharmacy*, 20<sup>th</sup> Edition, (2000), Lippincott Williams & Wilkins, Baltimore, MD.

Liquid form preparations include solutions, suspensions and emulsions. As an example may be mentioned water or water-propylene glycol solutions for parenteral injection or addition of sweeteners and opacifiers for oral solutions, suspensions and emulsions. Liquid form preparations may also include solutions for intranasal administration.

Aerosol preparations suitable for inhalation may include solutions and solids in powder form, which may be in combination with a pharmaceutically acceptable carrier, such as an inert compressed gas, e.g. nitrogen.

Also included are solid form preparations which are intended to be converted, shortly before use, to liquid form preparations for either oral or parenteral administration. Such liquid forms include solutions, suspensions and emulsions.

The compounds of the invention may also be deliverable transdermally. The transdermal compositions can take the form of creams, lotions, aerosols and/or emulsions and can be included in a transdermal patch of the matrix or reservoir type as are conventional in the art for this purpose.

Preferably the compound is administered orally.

Preferably, the pharmaceutical preparation is in a unit dosage form. In such form, the preparation is subdivided into suitably sized unit doses containing appropriate quantities of the active component, e.g., an effective amount to achieve the desired purpose.

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The quantity of active compound in a unit dose of preparation may be varied or adjusted from about 1 mg to about 350 mg, preferably from about 1 mg to about 150 mg, more preferably from about 1 mg to about 50 mg, according to the particular application.

The actual dosage employed may be varied depending upon the requirements of the patient and the severity of the condition being treated. Determination of the proper dosage regimen for a particular situation is within the skill of the art. For convenience, the total daily dosage may be divided and administered in portions during the day as required.

The amount and frequency of administration of the compounds of the invention and/or the pharmaceutically acceptable salts thereof will be regulated according to the judgment of the attending clinician considering such factors as age, condition and size of the patient as well as severity of the symptoms being treated. A typical recommended daily dosage regimen for oral administration can range from about 1 mg/day to about 300 mg/day, preferably 1 mg/day to 75 mg/day, in two to four divided doses.

When the invention comprises a combination of H<sub>3</sub> antagonist and H<sub>1</sub> antagonist compounds, the two active components may be co-administered simultaneously or sequentially, or a single pharmaceutical composition comprising a H<sub>3</sub> antagonist and an H<sub>1</sub> antagonist in a pharmaceutically acceptable carrier can be administered. The components of the combination can be administered individually or together in any conventional dosage form such as capsule, tablet, powder, cachet, suspension, solution, suppository, nasal spray, etc. The dosage of the H<sub>1</sub> antagonist can be determined from published material, and may range from 1 to 1000 mg per dose.

When separate H<sub>3</sub> and H<sub>1</sub> antagonist pharmaceutical compositions are to be administered, they can be provided in a kit comprising in a single package, one container comprising an H<sub>3</sub> antagonist in a pharmaceutically acceptable carrier, and a separate container comprising an H<sub>1</sub> antagonist in a pharmaceutically acceptable carrier, with the H<sub>3</sub> and H<sub>1</sub> antagonists being present in amounts such that the

combination is therapeutically effective. A kit is advantageous for administering a combination when, for example, the components must be administered at different time intervals or when they are in different dosage forms.

While the present invention has been described in conjunction with the specific embodiments set forth above, many alternatives, modifications and variations thereof will be apparent to those of ordinary skill in the art. All such alternatives, modifications and variations are intended to fall within the spirit and scope of the present invention.

### WHAT IS CLAIMED IS:

1. A compound represented by the structural formula

$$R^{1-X} \xrightarrow{(R^{12})_a} (R^{13})_b$$

$$R^{1-X} \xrightarrow{M^1} Y \xrightarrow{M^2} N Z$$

5 or a pharmaceutically acceptable salt or solvate thereof, wherein:

the dotted line represents an optional double bond;

a is 0 to 2;

b is 0 to 2;

n is 1, 2 or 3;

10 p is 1, 2 or 3;

r is 0, 1, 2, or 3;

with the provisos that when  $M^2$  is N, p is not 1; and that when r is 0,  $M^2$  is  $C(R^3)$ ; and that the sum of p and r is 1 to 4;

M<sup>1</sup> is C(R<sup>3</sup>) or N;

15  $M^2$  is  $C(R^3)$  or N;

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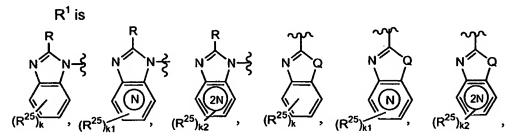
X is a bond or C<sub>1</sub>-C<sub>6</sub> alkylene;

Y is -C(O)-, -C(S)-,  $-(CH_2)_q$ -,  $-NR^4C(O)$ -,  $-C(O)NR^4$ -,  $-C(O)CH_2$ -,  $-SO_2$ -,  $-N(R^4)$ -, -NH-C(=N-CN)- or -C(=N-CN)-NH-; with the provisos that when M<sup>1</sup> is N, Y is not  $-NR^4C(O)$ - or -NH-C(=N-CN)-; when M<sup>2</sup> is N, Y is not  $-C(O)NR^4$ - or

20 –C(=N-CN)-NH-; and when Y is -N( $\mathbb{R}^4$ )-,  $\mathbb{M}^1$  is CH and  $\mathbb{M}^2$  is C( $\mathbb{R}^3$ );

q is 1 to 5, provided that when both M1 and M2 are N, q is 2 to 5;

Z is a bond,  $C_1$ - $C_6$  alkylene,  $C_1$ - $C_6$  alkenylene, -C(O)-, -CH(CN)-, -SO<sub>2</sub>- or -CH<sub>2</sub>C(O)NR<sup>4</sup>-;



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$$(R^{25})_{k}$$
,  $(R^{25})_{k1}$ ,  $(R^{25})_{k}$ ,  $(R^{25})_{k$ 

R is H, C<sub>1</sub>-C<sub>6</sub> alkyl, halo(C<sub>1</sub>-C<sub>6</sub>)alkyl-, C<sub>1</sub>-C<sub>6</sub> alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkyl-SO<sub>0-2</sub>, R<sup>32</sup>-aryl(C<sub>1</sub>-C<sub>6</sub>)alkoxy-, R<sup>32</sup>-aryl(C<sub>1</sub>-C<sub>6</sub>)alkyl-, R<sup>32</sup>-aryloxy, R<sup>32</sup>-heteroaryl, (C<sub>3</sub>-C<sub>6</sub>)cycloalkyl, (C<sub>3</sub>-C<sub>6</sub>)cycloalkyl-(C<sub>1</sub>-C<sub>6</sub>)alkyl, (C<sub>3</sub>-C<sub>6</sub>)cycloalkyl-(C<sub>1</sub>-C<sub>6</sub>)alkoxy, (C<sub>3</sub>-C<sub>6</sub>)cycloalkyl-oxy-, R<sup>37</sup>-heterocycloalkyl, R<sup>37</sup>-heterocycloalkyl-oxy-, R<sup>37</sup>-heterocycloalkyl-(C<sub>1</sub>-C<sub>6</sub>)alkoxy, N(R<sup>30</sup>)(R<sup>31</sup>)-(C<sub>1</sub>-C<sub>6</sub>)alkyl-, -N(R<sup>30</sup>)(R<sup>31</sup>), -NH-(C<sub>1</sub>-C<sub>6</sub>)alkyl-O-(C<sub>1</sub>-C<sub>6</sub>)alkyl, -NHC(O)NH(R<sup>29</sup>); R<sup>29</sup>-S(O)<sub>0-2</sub>-, halo(C<sub>1</sub>-C<sub>6</sub>)alkyl-S(O)<sub>0-2</sub>-, N(R<sup>30</sup>)(R<sup>31</sup>)-(C<sub>1</sub>-C<sub>6</sub>)alkyl-S(O)<sub>0-2</sub>- or benzoyl;

 $R^{8} \text{ is H, C}_{1}\text{-}C_{6} \text{ alkyl, halo}(C_{1}\text{-}C_{6}) \text{alkyl-, } (C_{1}\text{-}C_{6}) \text{alkoxy-}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{32}\text{-aryl}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{32}\text{-aryl, } R^{32}\text{-heteroaryl, } (C_{3}\text{-}C_{6}) \text{cycloalkyl, } (C_{3}\text{-}C_{6}) \text{cycloalkyl-}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{37}\text{-heterocycloalkyl, } N(R^{30})(R^{31})\text{-}(C_{1}\text{-}C_{6}) \text{alkyl-, } R^{29}\text{-}S(O)_{2}\text{-, halo}(C_{1}\text{-}C_{6}) \text{alkyl-}S(O)_{2}\text{-, } R^{29}\text{-}S(O)_{0}\text{-}_{1}\text{-}(C_{2}\text{-}C_{6}) \text{alkyl-, } halo(C_{1}\text{-}C_{6}) \text{alkyl-};}$ 

 $R^2$  is a six-membered heteroaryl ring having 1 or 2 heteroatoms independently selected from N or N-O, with the remaining ring atoms being carbon; a five-membered heteroaryl ring having 1, 2, 3 or 4 heteroatoms independently selected from N, O or S, with the remaining ring atoms being carbon;  $R^{32}$ -quinolyl;  $R^{32}$ -aryl; heterocycloalkyl;  $C_3$ - $C_6$ )cycloalkyl;  $C_1$ - $C_6$  alkyl; hydrogen; thianaphthenyl;

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wherein said six-membered heteroaryl ring or said five-membered heteroaryl ring is optionally substituted by R<sup>6</sup>;

R<sup>3</sup> is H, halogen, C<sub>1</sub>-C<sub>6</sub> alkyl, -OH, (C<sub>1</sub>-C<sub>6</sub>)alkoxy or -NHSO<sub>2</sub>-(C<sub>1</sub>-C<sub>6</sub>)alkyl;

 $R^4$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_6$  alkyl,  $C_3$ - $C_6$  cycloalkyl,  $(C_3$ - $C_6)$ cycloalkyl( $C_1$ - $C_6$ )alkyl,  $R^{33}$ -aryl,  $R^{33}$ -aryl( $C_1$ - $C_6$ )alkyl, and  $R^{32}$ -heteroaryl;

 $R^5$  is hydrogen,  $C_1$ - $C_6$  alkyl,  $-C(O)R^{20}$ ,  $-C(O)_2R^{20}$ ,  $-C(O)N(R^{20})_2$ ,  $(C_1$ - $C_6)$ alkyl- $SO_2$ -, or  $(C_1$ - $C_6)$ alkyl- $SO_2$ -NH-;

or R<sup>4</sup> and R<sup>5</sup>, together with the nitrogen to which they are attached, form an azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl or morpholinyl ring;

R<sup>6</sup> is 1 to 3 substituents independently selected from the group consisting of -OH, halogen, C<sub>1</sub>-C<sub>6</sub> alkyl-, C<sub>1</sub>-C<sub>6</sub> alkoxy, C<sub>1</sub>-C<sub>6</sub> alkylthio, -CF<sub>3</sub>, -NR<sup>4</sup>R<sup>5</sup>, -CH<sub>2</sub>-NR<sup>4</sup>R<sup>5</sup>, -NHSO<sub>2</sub>R<sup>22</sup>, -N(SO<sub>2</sub>R<sup>22</sup>)<sub>2</sub>, phenyl, R<sup>33</sup>-phenyl, NO<sub>2</sub>, -CO<sub>2</sub>R<sup>4</sup>, -CON(R<sup>4</sup>)<sub>2</sub>,

$$-NH-CH_2$$
 OCH<sub>3</sub> and  $-NH-CH_2$ 

 $R^7$  is  $-N(R^{29})$ -, -O- or  $-S(O)_{0-2}$ -;

 $R^{12}$  is independently selected from the group consisting of  $C_1$ - $C_6$  alkyl, hydroxyl,  $C_1$ - $C_6$  alkoxy, or fluoro, provided that when  $R^{12}$  is hydroxy or fluoro, then  $R^{12}$  is not bound to a carbon adjacent to a nitrogen; or two  $R^{12}$  substituents form a  $C_1$  to  $C_2$  alkyl bridge from one ring carbon to another non-adjacent ring carbon; or  $R^{12}$  is =0;

 $R^{13}$  is independently selected from the group consisting of  $C_1$ - $C_6$  alkyl, hydroxyl,  $C_1$ - $C_6$  alkoxy, or fluoro, provided that when  $R^{13}$  is hydroxy or fluoro then  $R^{13}$  is not bound to a carbon adjacent to a nitrogen; or two  $R^{13}$  substituents form a  $C_1$  to  $C_2$  alkyl bridge from one ring carbon to another non-adjacent ring carbon; or  $R^{13}$  is =0;

R<sup>20</sup> is independently selected from the group consisting of hydrogen, C<sub>1</sub>-C<sub>6</sub> alkyl, or aryl, wherein said aryl group is optionally substituted with from 1 to 3 groups independently selected from halogen, -CF<sub>3</sub>, -OCF<sub>3</sub>, hydroxyl, or methoxy; or when two R<sup>20</sup> groups are present, said two R<sup>20</sup> groups taken together with the nitrogen to which they are bound can form a five or six membered heterocyclic ring;

R<sup>22</sup> is C<sub>1</sub>-C<sub>6</sub> alkyl, R<sup>34</sup>-aryl or heterocycloalkyl;

 $R^{24}$  is H,  $C_1$ - $C_6$  alkyl, -SO<sub>2</sub> $R^{22}$  or  $R^{34}$ -aryl;

 $R^{25}$  is independently selected from the group consisting of  $C_1$ - $C_6$  alkyl, halogen, -CN, -NO<sub>2</sub>, -CF<sub>3</sub>, -OH, C<sub>1</sub>-C<sub>6</sub> alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkyl-C(O)-, aryl-C(O)-, -C(O)OR<sup>29</sup>,

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-N(R<sup>4</sup>)(R<sup>5</sup>), N(R<sup>4</sup>)(R<sup>5</sup>)-C(O)-, N(R<sup>4</sup>)(R<sup>5</sup>)-S(O)<sub>1-2</sub>-, R<sup>22</sup>-S(O)<sub>0-2</sub>-, halo-(C<sub>1</sub>-C<sub>6</sub>)alkyl- or halo-(C<sub>1</sub>-C<sub>6</sub>)alkoxy-(C<sub>1</sub>-C<sub>6</sub>)alkyl-;

 $R^{29}$  is H,  $C_1$ - $C_6$  alkyl,  $C_3$ - $C_6$  cycloalkyl,  $R^{35}$ -aryl or  $R^{35}$ -aryl ( $C_1$ - $C_6$ )alkyl-;  $R^{30}$  is H,  $C_1$ - $C_6$  alkyl-,  $R^{35}$ -aryl or  $R^{35}$ -aryl ( $C_1$ - $C_6$ )alkyl-;

$$\begin{split} R^{31} \text{ is H, C}_{1}\text{-}C_{6} \text{ alkyl-, R}^{35}\text{-}\text{aryl, R}^{35}\text{-}\text{aryl(C}_{1}\text{-}C_{6})\text{alkyl-, R}^{35}\text{-}\text{heteroaryl, (C}_{1}\text{-}C_{6})\text{alkyl-C(O)-, R}^{35}\text{-}\text{aryl-C(O)-, N(R}^{4})\text{(R}^{5})\text{-}\text{C(O)-, (C}_{1}\text{-}C_{6})\text{alkyl-S(O)}_{2}\text{-}\text{ or R}^{35}\text{-}\text{aryl-S(O)}_{2}\text{-};\\ \text{or R}^{30} \text{ and R}^{31} \text{ together are -(CH}_{2})_{4\text{-}5}\text{-}, -(CH}_{2})_{2}\text{-}\text{O-(CH}_{2})_{2}\text{-}\text{ or } \end{split}$$

-(CH<sub>2</sub>)<sub>2</sub>-N(R<sup>38</sup>)-(CH<sub>2</sub>)<sub>2</sub>- and form a ring with the nitrogen to which they are attached; R<sup>32</sup> is 1 to 3 substituents independently selected from the group consisting of

H, -OH, halogen,  $C_1$ - $C_6$  alkyl,  $C_1$ - $C_6$  alkoxy,  $R^{35}$ -aryl-O-, -SR<sup>22</sup>, -CF<sub>3</sub>, -OCF<sub>3</sub>, -OCHF<sub>2</sub>, -NR<sup>39</sup>R<sup>40</sup>, phenyl,  $R^{33}$ -phenyl, NO<sub>2</sub>, -CO<sub>2</sub>R<sup>39</sup>, -CON( $R^{39}$ )<sub>2</sub>, -S(O)<sub>2</sub>R<sup>22</sup>, -S(O)<sub>2</sub>N( $R^{20}$ )<sub>2</sub>, -N( $R^{24}$ )S(O)<sub>2</sub>R<sup>22</sup>, -CN, hydroxy-(C<sub>1</sub>-C<sub>6</sub>)alkyl-, -OCH<sub>2</sub>CH<sub>2</sub>OR<sup>22</sup>, and  $R^{35}$ -aryl(C<sub>1</sub>-C<sub>6</sub>)alkyl-O-, or two  $R^{32}$  groups on adjacent carbon atoms together form a -OCH<sub>2</sub>O- or -O(CH<sub>2</sub>)<sub>2</sub>O- group;

R<sup>33</sup> is 1 to 3 substituents independently selected from the group consisting of C<sub>1</sub>-C<sub>6</sub> alkyl, halogen, -CN, -NO<sub>2</sub>, -CF<sub>3</sub>, -OCF<sub>3</sub>, -OCHF<sub>2</sub> and -O-(C<sub>1</sub>-C<sub>6</sub>)alkyl;

 $R^{34}$  is 1 to 3 substituents independently selected from the group consisting of H, halogen, -CF<sub>3</sub>, -OCF<sub>3</sub>, -OH and -OCH<sub>3</sub>;

 $R^{35}$  is 1 to 3 substituents independently selected from hydrogen, halo,  $C_1$ - $C_6$  alkyl, hydroxy,  $C_1$ - $C_6$  alkoxy, phenoxy, -CF<sub>3</sub>, -N( $R^{36}$ )<sub>2</sub>, -COOR<sup>20</sup> and -NO<sub>2</sub>;

 $R^{36}$  is independently selected form the group consisting of H and  $C_1$ - $C_6$  alkyl;  $R^{37}$  is 1 to 3 substituents independently selected from hydrogen, halo,  $C_1$ - $C_6$  alkyl, hydroxy,  $C_1$ - $C_6$  alkoxy, phenoxy, -CF<sub>3</sub>, -N( $R^{36}$ )<sub>2</sub>, -COOR<sup>20</sup>, -C(O)N( $R^{29}$ )<sub>2</sub> and -NO<sub>2</sub>, or  $R^{37}$  is one or two =O groups;

 $R^{38}$  is H,  $C_1$ - $C_6$  alkyl,  $R^{35}$ -aryl,  $R^{35}$ -aryl( $C_1$ - $C_6$ )alkyl-, ( $C_1$ - $C_6$ )alkyl-SO<sub>2</sub> or halo( $C_1$ - $C_6$ )alkyl-SO<sub>2</sub>-;

 $R^{39}$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_6$  alkyl,  $C_3$ - $C_6$  cycloalkyl,  $(C_3$ - $C_6)$ cycloalkyl( $C_1$ - $C_6)$ alkyl,  $R^{33}$ -aryl,  $R^{33}$ -aryl( $C_1$ - $C_6)$ alkyl, and  $R^{32}$ -heteroaryl; and

 $R^{40}$  is hydrogen,  $C_1$ - $C_6$  alkyl,  $-C(O)R^{20}$ ,  $-C(O)_2R^{20}$ ,  $-C(O)N(R^{20})_2$ ,  $(C_1$ - $C_6$ )alkyl- $SO_2$ -, or  $(C_1$ - $C_6$ )alkyl- $SO_2$ -NH-;

or R<sup>39</sup> and R<sup>40</sup>, together with the nitrogen to which they are attached, form an azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl or morpholinyl ring.

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- 2. A compound of claim 1 wherein M<sup>1</sup> is N, a is 0, and n is 2, and the optional double bond is not present.
- 3. A compound of claim 2 wherein M<sup>2</sup> is C(R<sup>3</sup>) wherein R<sup>3</sup> is hydrogen or fluorine, 5 b is 0, r is 1, and p is 2.
  - 4. A compound of claim 3 wherein X is a bond.
  - 5. A compound of claim 4 wherein Y is -C(O)-.
  - 6. A compound of claim 5 wherein Z is straight or branched C<sub>1</sub>-C<sub>3</sub> alkyl.
    - 7. A compound of claim 6 wherein  $R^2$  is a six-membered heteroaryl ring, optionally substituted with one  $R^6$  substituent.
  - 8. A compound of claim 7 wherein R<sup>2</sup> is pyrimidy, R<sup>6</sup>-pyrimidyl, pyridyl, R<sup>6</sup>-pyridyl or pyridazinyl and R<sup>6</sup> is -NH<sub>2</sub>.
    - 9. A compound of claim 8 wherein R<sup>2</sup> is

$$\begin{cases} N & \begin{cases} N \\ N \end{cases} & N \end{cases}$$
 or 
$$\begin{cases} N \\ N \end{cases}$$

10. A compound of claim 2 wherein R<sup>1</sup> is

$$\mathbb{R}^{\mathbb{N}}$$
  $\mathbb{N}^{-\xi}$   $\mathbb{N}^{\mathbb{N}^{-\xi}}$  or  $\mathbb{R}^{\mathbb{N}^{25}}$ 

- 25 11. A compound of claim 10 wherein R is (C<sub>1</sub>-C<sub>6</sub>)alkyl, (C<sub>1</sub>-C<sub>6</sub>)alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkoxy, (C<sub>1</sub>-C<sub>6</sub>)alkylthio, heteroaryl or R<sup>32</sup>-aryl; R<sup>25</sup> is halogen or CF<sub>3</sub>; and k and k1 are 0 or 1.
- 12. A compound of claim 11 wherein R is -CH<sub>3</sub>, -CH<sub>2</sub>CH<sub>3</sub>, -OCH<sub>3</sub>, -OCH<sub>2</sub>CH<sub>3</sub>, 30 OCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, -OCH((CH<sub>3</sub>)<sub>2</sub>, -CH<sub>2</sub>CH<sub>3</sub>, -SCH<sub>3</sub>, -SCH<sub>2</sub>CH<sub>3</sub>, pyridyl, pyrimidyl, pyrazinyl, furanyl, oxazolyl or R<sup>32</sup>-phenyl.

A compound of claim 12 wherein  $\mathbb{R}^2$  is 13.

$$\begin{cases} N_{NH_2} & \begin{cases} N_{NH_2} & \\ N_{NH_2} & \end{cases} \end{cases} \end{cases} \end{cases}$$

A compound of claim 1 selected from the group consisting of 14.

- 15. A pharmaceutical composition comprising an effective amount of a compound of claim 1 and a pharmaceutically effective carrier.
- 16. The use of a compound of claim 1 for the preparation of a medicament for treating allergy, allergy-induced airway responses, congestion, hypotension, cardiovascular disease, diseases of the GI tract, hyper and hypo motility and acidic secretion of the gastro-intestinal tract, obesity, sleeping disorders, disturbances of the central nervous system, attention deficit hyperactivity disorder, hypo and hyperactivity of the central nervous system, Alzheimer's disease, schizophrenia, and migraine.

17. A pharmaceutical composition comprising an effective amount of a compound of claim 1, and an effective amount of H<sub>1</sub> receptor antagonist, and a pharmaceutically effective carrier.

- 18. The use of a compound of claim 1 for the preparation of a medicament for use in combination with an H<sub>1</sub> receptor antagonist for treating allergy, allergy-induced airway responses, and congestion.
- 19. The use of claim 19 wherein said H<sub>1</sub> receptor antagonist is selected from:
  20 astemizole, azatadine, azelastine, acrivastine, brompheniramine, cetirizine, chlorpheniramine, clemastine, cyclizine, carebastine, cyproheptadine, carbinoxamine, descarboethoxyloratadine, diphenhydramine, doxylamine, dimethindene, ebastine, epinastine, efletirizine, fexofenadine, hydroxyzine, ketotifen, loratadine, levocabastine, meclizine, mizolastine, mequitazine, mianserin, noberastine, norastemizole, picumast, pyrilamine, promethazine, terfenadine, tripelennamine, temelastine, trimeprazine or triprolidine.

Internat Application No PCT/US 03/11672

A. CLASSIFICATION OF SUBJECT MATTER IPC 7 A61K31/4545 C07D471/04 CO7D401/14 A61K31/4184 A61P37/08 According to International Patent Classification (IPC) or to both national classification and IPC B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC 7 CO7D Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) EPO-Internal, WPI Data, BEILSTEIN Data, CHEM ABS Data C. DOCUMENTS CONSIDERED TO BE RELEVANT Relevant to claim No. Citation of document, with Indication, where appropriate, of the relevant passages Category \* Y US 6 211 199 B1 (KUDLACZ ELIZABETH M ET 1-19 AL) 3 April 2001 (2001-04-03) claims; examples EP 0 580 541 A (ESPANOLA PROD QUIMICOS) Y 1-19 26 January 1994 (1994-01-26) the whole document -/--Further documents are listed in the continuation of box C. Patent family members are listed in annex. X Special categories of cited documents: "T" later document published after the international filing date or priority date and not in conflict with the application but "A" document defining the general state of the art which is not considered to be of particular relevance cited to understand the principte or theory underlying the invention 'E' earlier document but published on or after the international "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone 'L' document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled "O" document referring to an oral disclosure, use, exhibition or \*P\* document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 1 August 2003 18/08/2003 Name and mailing address of the ISA Authorized officer European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo ni, Fec (+31-70) 340-3016 Gavriliu, D

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		PCT/US 03/11672		
	ation) DOCUMENTS CONSIDERED TO BE RELEVANT			
Category *	Citation of document, with Indication, where appropriate, of the relevant passages	Relevant to ctalm No.		
Y	JANSSENS F ET AL: "NEW ANTIHISTAMINIC N-HETEROCYCLIC 4-PIPERIDINAMINES. 3. SYNTHESIS AND ANTIHISTAMINIC ACTIVITY OF N-(4-PIPERIDINYL)-3H-IMIDAZO4,5-BPYRIDIN-2-AMINES"  JOURNAL OF MEDICINAL CHEMISTRY, AMERICAN CHEMICAL SOCIETY. WASHINGTON, US, vol. 28, no. 12, 1985, pages 1943-1947, XP001084054 ISSN: 0022-2623 table II	1-19		
A	WO 02 24659 A (SCHERING CORP) 28 March 2002 (2002-03-28) claims 1,5,14-23	1-19		
A	EP 0 626 373 A (SUMITOMO PHARMA) 30 November 1994 (1994-11-30) page 20, line 1 -page 31, line 10; claims; examples 15,17-19,23-25,28,46-48 page 55, line 25 -page 58, line 55	1-19		

# FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Claims Nos.: 1-4(partially); 15-19(partially)

Present claims 1-3, 15-19 relate to an extremely large number of possible compounds, compositions and uses. Support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT is to be found, however, for only a very small proportion of the compounds, compositions and uses claimed. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Consequently, the search has been carried out for those parts of the claims which appear to be supported and disclosed, namely those parts relating to the compounds of Formula I, in which X is a bond or C1 alkyl; Y is C(0), C(S), CH2, SO2, N(R4); M1 is N, a=0, n=2, the optional double bond is not present, r=1 and p=2. The search is complete for all the examples of the description.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

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Box I O	bservations where certain claims were found unsearchable (Continuation of Item 1 of first sheet)
This interna	ational Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
	aims Nos.: ecause they relate to subject matter not required to be searched by this Authority, namely:
be ar	alms Nos.:  1-4(partially); 15-19(partially) ecause they relate to parts of the International Application that do not comply with the prescribed requirements to such a extent that no meaningful International Search can be carried out, specifically:  ee FURTHER INFORMATION sheet PCT/ISA/210
	alms Nos.: scause they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).
Box Ii O	bservations where unity of invention is lacking (Continuation of item 2 of first sheet)
This Interna	ational Searching Authority found multiple inventions in this international application, as follows:
	s all required additional search fees were timely paid by the applicant, this international Search Report covers all tearchable claims.
	s all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment any additional fee.
	s only some of the required additional search fees were timely paid by the applicant, this International Search Report overs only those claims for which fees were paid, specifically claims Nos.:
4. N	o required additional search fees were timely paid by the applicant. Consequently, this International Search Report is stricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark or	Protest  The additional search fees were accompanied by the applicant's protest.  No protest accompanied the payment of additional search fees.

Internation Application No
PCT/US 03/11672

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